

Perspectives of Molecular Manipulation and Fabrication

September 28, 2016 | F. S. Tautz

Peter Grünberg Institut, Forschungszentrum Jülich

Workshop I, Machnine Learning Meets Many-Particle Problems, Long Program "Understanding Many-Particle Systems with Machine Learning", Institute for Pure and Applied Mathematics, UCLA Feynman's Vision



"There is plenty of room at the bottom" (R. Feynman, APS meeting, Pasadena, December 1959)

 \rightarrow inspired the nanoscience revolution

Nanoscience in broad sense: Every activity involving structures smaller than 100 nm

But: Feynman had a more specific vision: miniaturized manufacturing and fabrication technology in analogy with macroscopic world, but ultimately at the atomic scale.





A universal molecular assembly machine

"3D printing" of molecular structures at the single molecule level.





The Prize



Making and harnessing artificial molecular nanostructures

- There is a lot of activity in this direction...
- Study biological examples
- Develop concepts of artificial functional nanostrucures: Inspiration by nature + physical and chemical intuition + simulations
- Design and synthesis of appropriate molecules

>But: How to assemble the molecular building blocks?

Self-assembly ? Directed manufacture !

Where Do We Stand?



Scanning Tunneling Microscopy Binnig and Rohrer, Nobel Prize in Physics in 1984



B. Voigtländer, Forschungszentrum Jülich



Where Do We Stand?



Scanning Tunneling Microscopy Binnig and Rohrer, Nobel Prize in Physics in 1984



Where Do We Stand?



Scanning Tunneling Microscopy Binnig and Rohrer, Nobel Prize in Physics in 1984

STM @ Low temperature: Immobilize single atoms and molecules (Xe on Nickel) Eigler and Schweizer 1990



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SPM and Molecular Fabrication



- Cannot work atom-by-atom: Energy released when covalent bonds form is disruptive in weakly bonded structures
 - \rightarrow work with molecules
- Cannot use push, pull, pick-up transfer drop approaches, because molecules are anisotropic and floppy (many internal degrees of freedom).

 \rightarrow need a more controlled approach

"Complexity Gap" (information gap, control gap) Many degrees of freedom to be measured / controlled vs.

Two quantities (force, conductance) that can be measured and three degrees of freedom that can be controlled (tip)





Determining van der Waals Coefficients

Molecular Manipulation Lab (MOMALAB)

The Connection to Machine Learning





Since Eigler, manipulation techniques have not changed **push, pull, pick-up – transfer – drop**

(Logic gates, movie, ...)

Drawbacks:

- Essentially, these are stochastic techniques: During key steps of manipulation, the control is lost
- This presents a serious limitation for molecular (as opposed to atomic) manipulation



At any time, **two contacts** to the entity to be manipulated are maintained.

- Fully deterministic (at least in principle)
- Internal degrees of freedom become addressable





Desirables:

- As many information channels about the state of the junction
 - \rightarrow simultaneous STM and AFM (current and force)

Pre-requisites:

- Ultra-high vacuum
- Low temperature
- Stiff cantilevers: The force measuring unit must be able to counter all force gradients that arise during the experiment. (otherwise: temporary loss of control)

Dynamic Non-Contact Atomic Force Microscope







Dynamic Non-Contact Atomic Force Microscope





Dynamic Non-Contact Atomic Force Microscope









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Individual $\Delta f(z)$ "spectra"



Continuous – Reproducible – Reversible



Histograms



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Qualitative Understanding

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Inverse Problem

$$\mathbf{R}_{1} \dots \mathbf{R}_{M}$$
Inverse Problem
$$\mathbf{R}_{1} \dots \mathbf{R}_{M}$$
Inverse Problem
$$\mathbf{R}_{1} \dots \mathbf{R}_{M}$$
Inverse Problem

Configuration of the molecule in the junction

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Classical, Quasistatic Simulation



$$V = V_{\text{intra}} + V_{\text{mol-surf}} + V_{\text{mol-tip}}$$



Molecular mechanics ("force field")

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Classical, Quasistatic Simulation





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Van der Waals Potential



Asymptotic Potential



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Lifshitz-Zaremba-Kohn



Experiment





Experimental Results: NTCDA, PTCDA, TTCDA





Experimental Results: NTCDA, PTCDA, TTCDA





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Fitting the Experimental Data



$$\Delta f_{[\mathrm{N},\mathrm{P},\mathrm{T}]}(z_{\mathrm{tip}}) = \frac{d^2 V}{dz_{\mathrm{tip}}^2}$$
$$V = -\sum_{i=1}^{M} \frac{\gamma_i C_{\alpha,[\mathrm{N},\mathrm{P},\mathrm{T}]}}{((z_{\mathrm{tip}} - d_i) - z_0)^{\alpha}}$$

Fit parameters α , $C_{\alpha,N}$, $C_{\alpha,P}$, $C_{\alpha,T}$, z_0

NTCDA, PTCDA, TTCDA are fitted simultaneously

Individual C-coefficients for each molecule (beyond additivity)

Atomic weighting factors from V. G. Ruiz et al., PRL 2012:

 $\gamma_{\rm H} = 0.29$ $\gamma_{\rm O} = 0.67$ $\gamma_{\rm C} = 1.00$

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Step 1: Fitting the Asymptotic Force Law



 $\alpha \approx 3$ (vdW interaction dominates the molecule-surface potential)

 $z_0 \approx d_{Au(111)}/2$ (correct distance calibration)

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Step 2: Fitting C_3 Coefficients and z_0 (with $\alpha \equiv 3$)





Asymptotic α =3 behaviour for z_{mol} > 5 Å

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 C_3 -coefficients converge for z_{mol} > 5 Å, to different values for the three molecules





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Superlinear Rise of the Molecular C₃ Coefficients



Carbon atoms in TTCDA are **more polarizable** than in NTCDA,

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i.e. the van der Waals interaction is **not additive** over the carbon atoms.

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Superlinear Rise of the Molecular C₃ Coefficients



DFT + **semi-empirical vdW**^{surf}:

(=isotropic free-atom static polarizability, volume-scaled for immediate chemical environment).

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by construction all carbon atoms are the same

(vdW^{surf}: Ruiz et al PRL 2012)

Superlinear Rise of the Molecular C₃ Coefficients





DFT + random phase approximation:

(=dynamic linear response theory, self-consistent but linearized Hartree calculation of electron density in the molecule in presence of external perturbation and its screening in molecule).

The same superlinearity as in experiment is observed.



Origin of Superlinearity

Microscopic response function of the molecule: 1st order perturbation of wave functions



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Optical Polarizability vs. van der Waals Force





Why is the superlinear effect in the van der Waals interaction so small?

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Optical Polarizability vs. van der Waals Force





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Short Range Potential





Short Range Potential





Fit Result





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Phys. Rev. Lett. 109, 076102 (2012)

Consistency btw. Asymptotic and Short Range





 $C_3^{eff} \approx 70 \text{ kcal} / \text{mol} \text{ Å}^3$

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Seamless Matching

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Seamless Matching





Adsorption Energy





PTCDA/Au(111)

Single molecule adsorption energy of 2.5 eV

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Dissecting the Bonding Type





Pure van der Waals bond, no chemical interaction Dispersion interaction $V_{disp} \approx 90 \text{ meV}$ / atom



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$$V(\vec{R}_1 \dots \vec{R}_M; \vec{\mathcal{R}}_1 \dots \vec{\mathcal{R}}_P) = \sum_{i,j} V_{i,j}^{\text{bond}} + \sum_{i,j,k} V_{i,j,k}^{\text{angle}} + \sum_{i,j,k,l} V_{i,j,k,l}^{\text{dihedral}} + \sum_{i,j,k,l} V_{i,j,$$

Molecular mechanics ("force field")



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$$+ \sum_{i,j} V_{i,j}^{\text{nb}} + \sum_i V_i^{\text{a-s}} + \sum_i V_i^{\text{corr}} + V_{\text{mol-tip}},$$

Molecular mechanics ("force field")







Surface Corrugation: Apparent Repulsive Force





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Surface Corrugation: Simulation





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Comprehensive Simulation





Summary

- "Molecules with handle" + STM/AFM:
 Highly reproducible manipulation experiments
- ✓ van der Waals energy of a single molecule on a metal can be measured



asymptotic exponent α =3, reference plane position $z_0 \approx 1.1$ Å, validity range z>5 Å, absolute size of C₃ coefficients $\approx 25-28$ kcal/mol Å³, $\approx 10\%$ superlinearity (i.e. non-additivity)

- ✓ Full surface holding potential can be reconstructed ($z_0 \approx 0$, $C_3 \approx 70$ kcal/mol Å³)
- ✓ Adsorption energy can be determined ($\approx 2.5 \text{ eV}$)
- ✓ Bonding energy can be partitioned (≈90 meV/atom)
- ✓ Lateral corrugation introduces apparent repulsive force and jerky sliding
- Force-field simulations describe the manipulation process reasonably accurate







Molecular Manipulation Lab



How to develop two-contact manipulation into a fabrication technique?





drop the restriction to vertical (and rectilinear) tip motion: curvilinear tip motion

- enable playful, intuitive handling of molecules, by introducing hand-controlled manipulation
- obtain complementary information about the state of the junction from simulations

Ink experiments and simulations in real time

Wagner, Temirrov Tautz, in "Molecular Architectonics" Ed. T Ogawa, Springer 2017

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Molecular Manipulation Laboratory





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Hand controlled manipulation

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Virtual Reality





- Offline data visualization
- Guiding and transfering hand-controlled manipulation protocols
- In conjunction with real-time simulation: The operator sees at the atomic scale what he is doing while doing it

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Outlook

Lack of Robustness



(residual stochasticity, not every tip works etc.)



thermal, quantum mechanical?



Is there a **quantum mechanical limit** to controlling the position of the molecule during manipulation?

(i) The molecule might **tunnel out of a potential minimum**:

$$\exp(-d\sqrt{2m(V_0-E)}/\hbar)$$

→ negligible because of the molecule's large mass.
 (ii) Because of zero point motion, the position of the molecule is indeterminate.

We have $\hbar \omega > k_{\rm B}T ~(T \simeq 10 \, {\rm K}) ~(k \simeq 25 \, {\rm N/m})$

 \rightarrow the molecule behaves as a quantum oscillator.

But:
$$\Delta x = \sqrt{\hbar/(2m\omega)} \simeq 0.04 \text{ Å}$$

Wagner, Temirrov Tautz, in "Molecular Architectonics" Ed. T Ogawa, Springer 2017

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Information Gap



Lack of information about the state of the junction



 configurations a,b,c, are specified by {R₁, ... R_M},
 each corresponds to a local minimum in the PES over 3Mdimensional configuration space E({R₁, ... R_M}) Coming Back to Feynman's Vision...



...of being able to make everything which can possibly be made, molecule by molecule:

Even if this will never replace self-assembly as bread-andbutter approach to nanotechnology,
being able to make a prototype of every conceivable molecular nanostructure gives us unlimited freedom to study emergent phenomena at the quantum scale

"...might tell us much of great interest about the phenomena that occur in complex situations"



Christian Wagner



Matthew Green



Ruslan Temirov



Philip Leinen



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