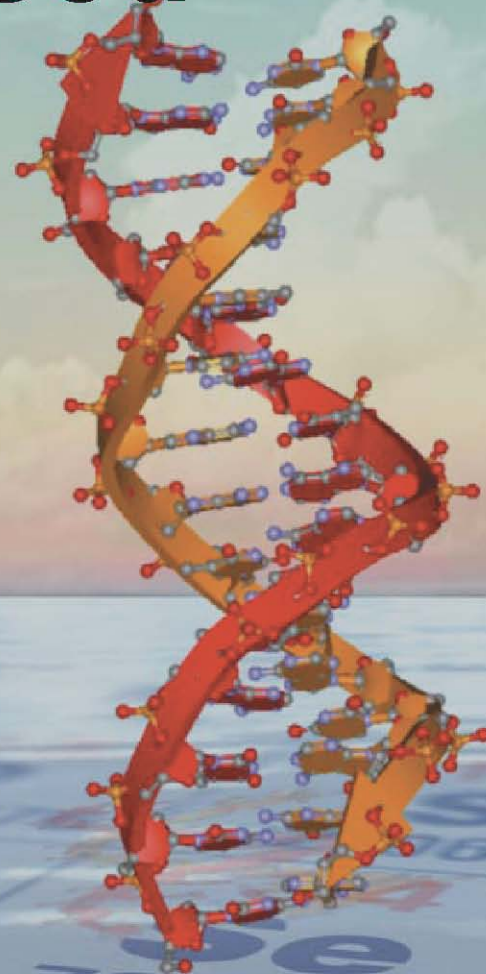
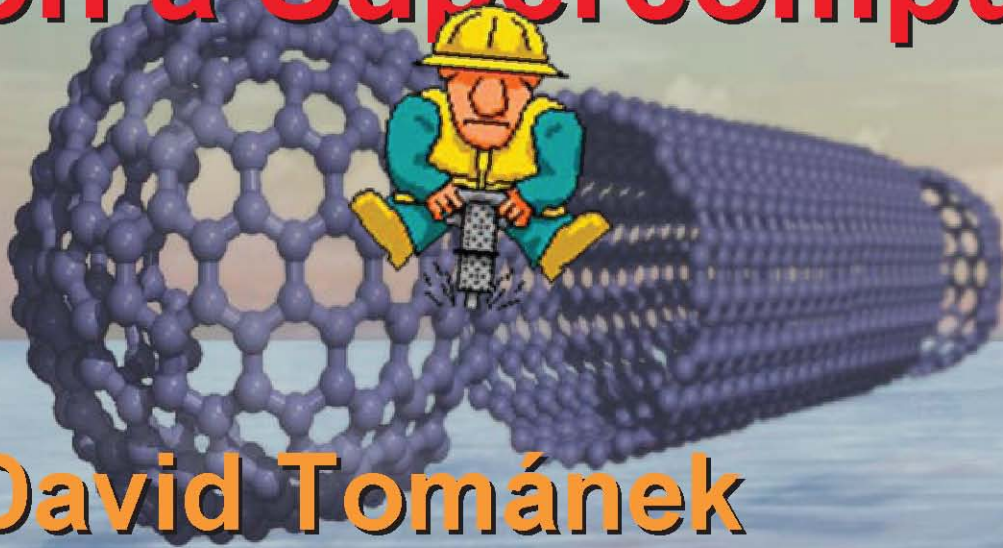


Designing Carbon-Based Nanotechnology on a Supercomputer



David Tománek

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Acknowledgements

Savas Berber,	<i>University of Tsukuba, Japan</i>
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Mina Yoon,	<i>Oak Ridge National Laboratory</i>

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NSF-NIRT

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RIST (Japan)

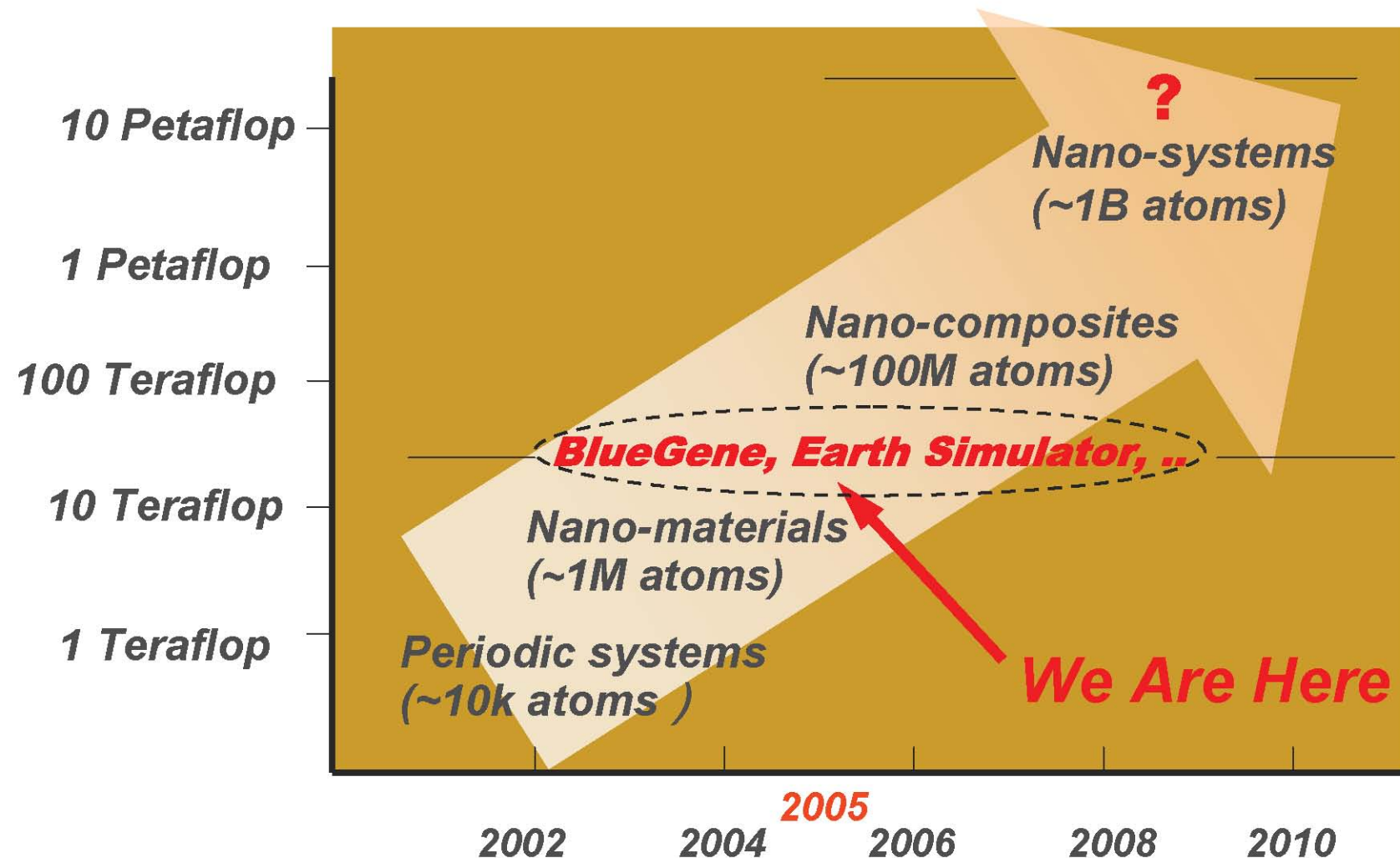


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David Tománek, Carbon-based nanotechnology on a supercomputer, Topical Review in J. Phys.: Condens. Matter 17, R413-R459 (2005).

State of the Art Computer Simulations: Where Are We Now?



The Nanocarbon Laboratory: Earth Simulator, Japan

www.nytimes.com

The New York Times
— ON THE WEB —

April 20, 2002

Japanese Computer Is World's Fastest, as U.S. Falls Back

By JOHN MARKOFF

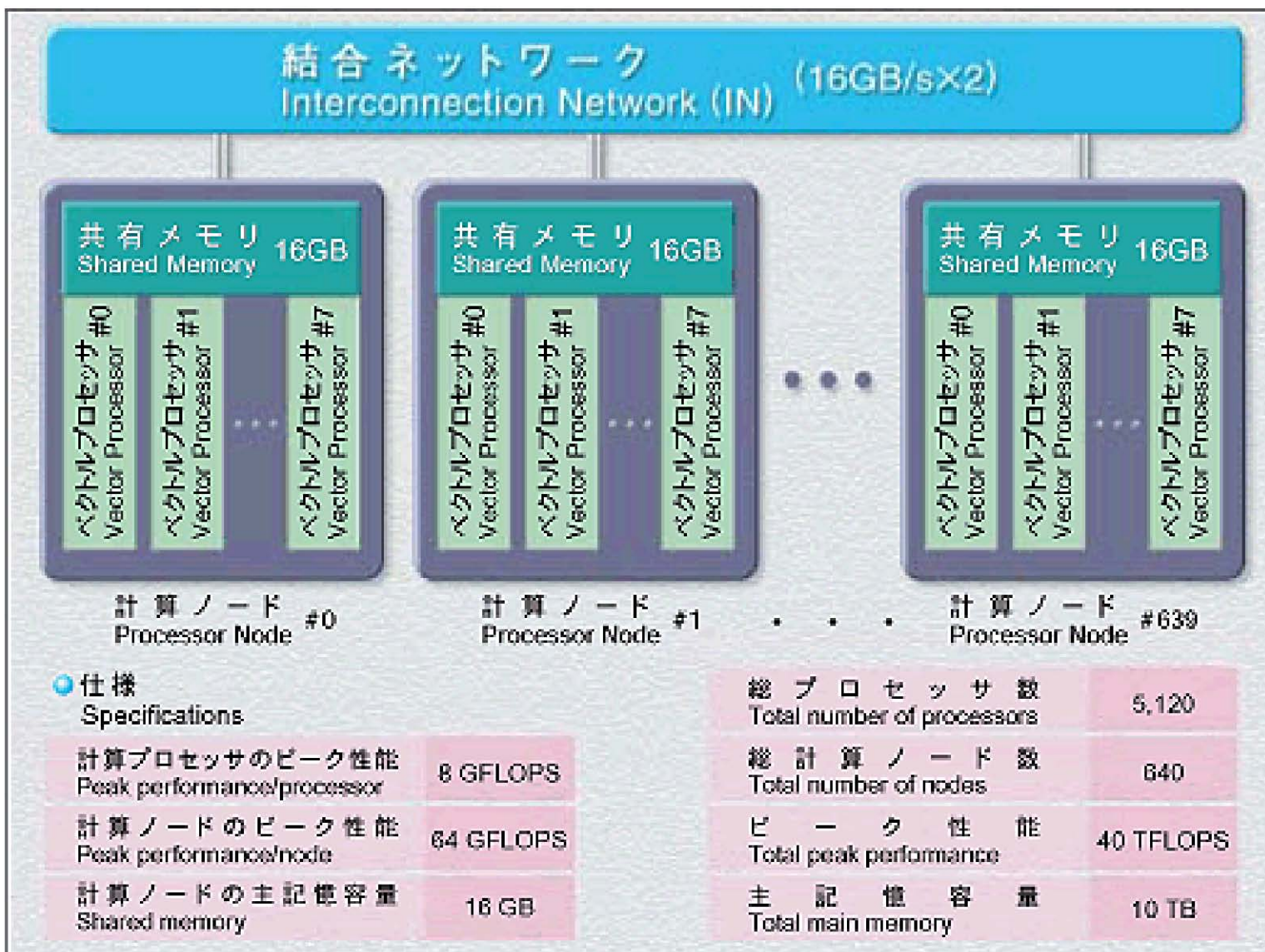
SAN FRANCISCO, April 19 — A Japanese laboratory has built the world's fastest computer, a machine so powerful that it matches the raw processing power of the 20 fastest American computers combined and far outstrips the previous leader, an I.B.M.-built machine.

Earth Simulator Supercomputer

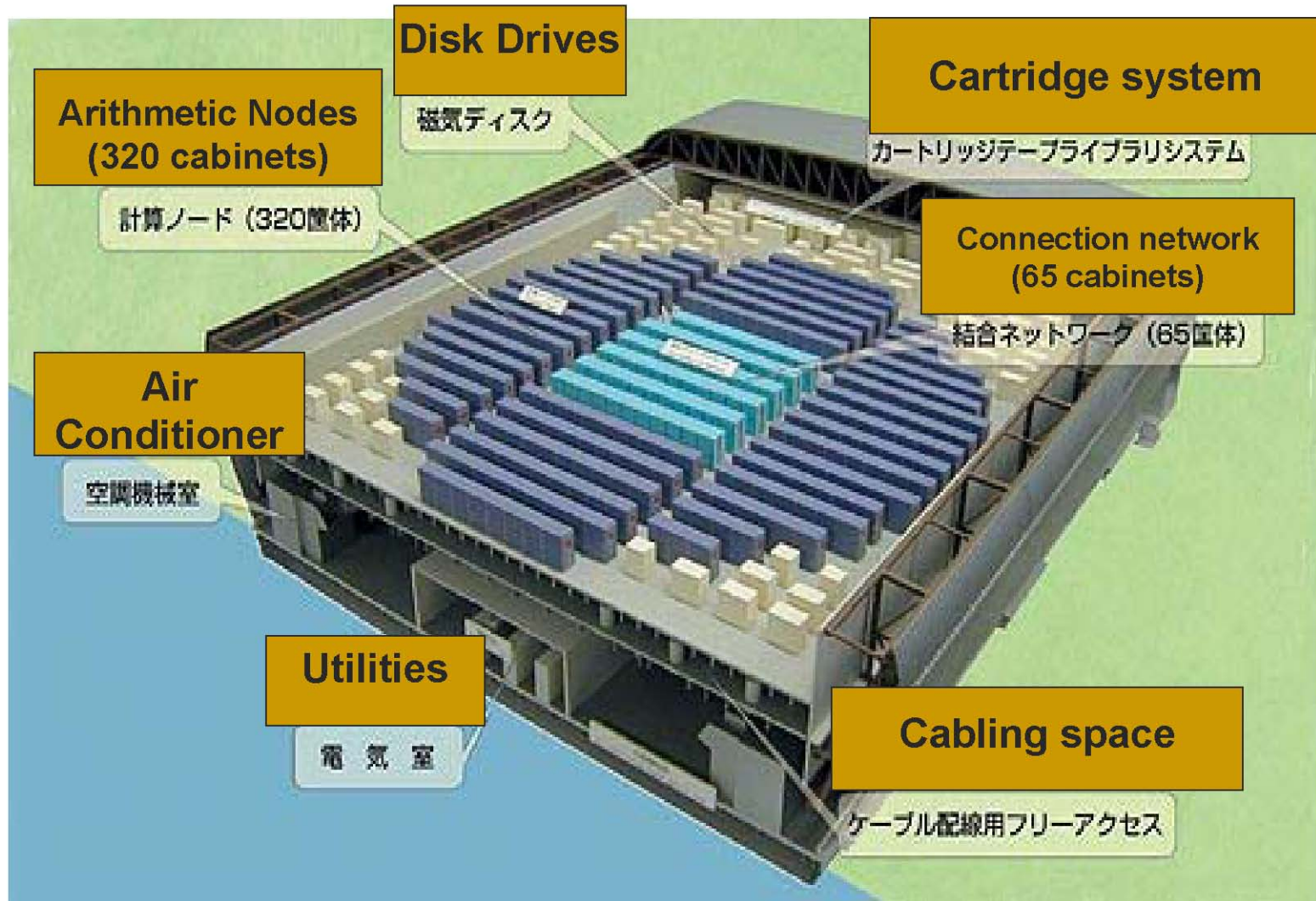
- Designed to model the Solid Earth (climate, earthquakes)
- Construction: 1997-2002
- Operation: since late 2002, by JAMSTEC
- Cost: 500M\$
- Vendor: NEC
- Type:
 - Massively vector parallel architecture
 - 5120 processors/ 640 nodes
 - 40 TFLOPS, 10 TB
 - OS: Unix, MPI / OpenMP, FORTRAN77/90, C
- Location: Yokohama



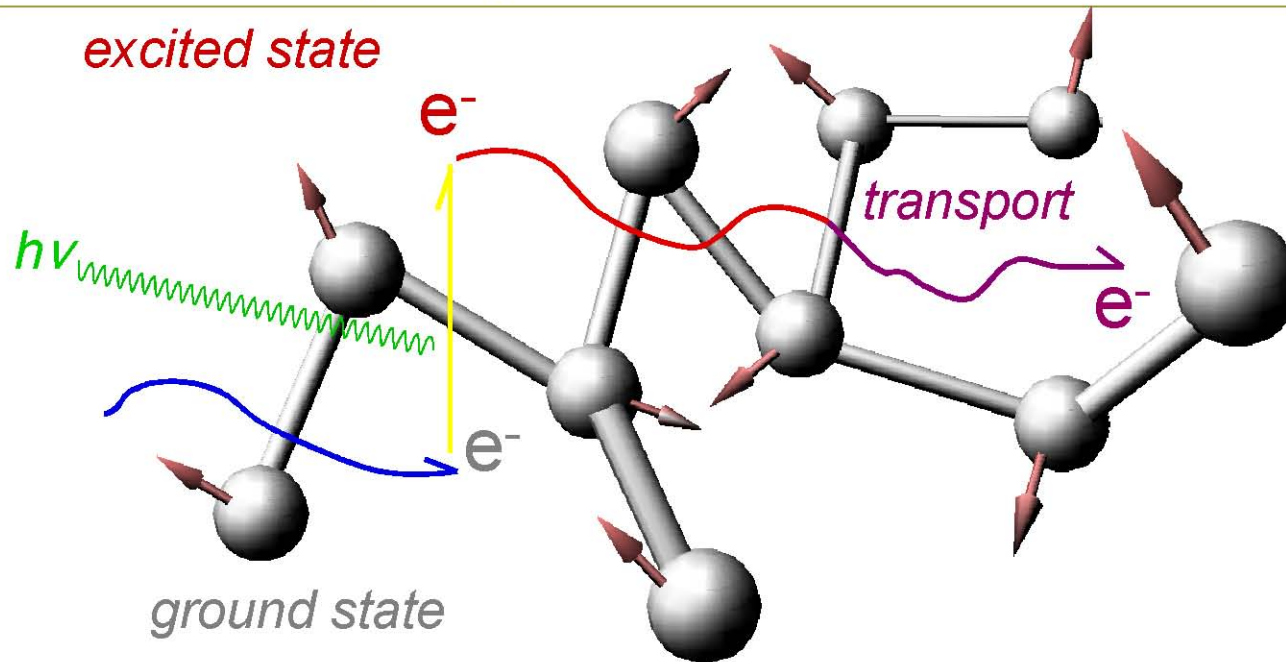
Earth Simulator Architecture



Earth Simulator Layout



Challenging Problems: Microscopic processes in nanostructures



Which processes follow exposure to

- Heat?
- Photo-excitations?
- Carrier injection?
- Atomic collisions?

Computational Approaches

- Electronic structure calculations based on the *ab initio* Density Functional formalism

- Atomic motion in the electronic ground state: Molecular dynamics simulations

- Forces from total energy expressions:

$$E_{\text{tot}} = E_{\text{tot}}(\{R_i\}) = E_{\text{tot}}\{\rho(r)\}$$

ab initio Density Functional formalism

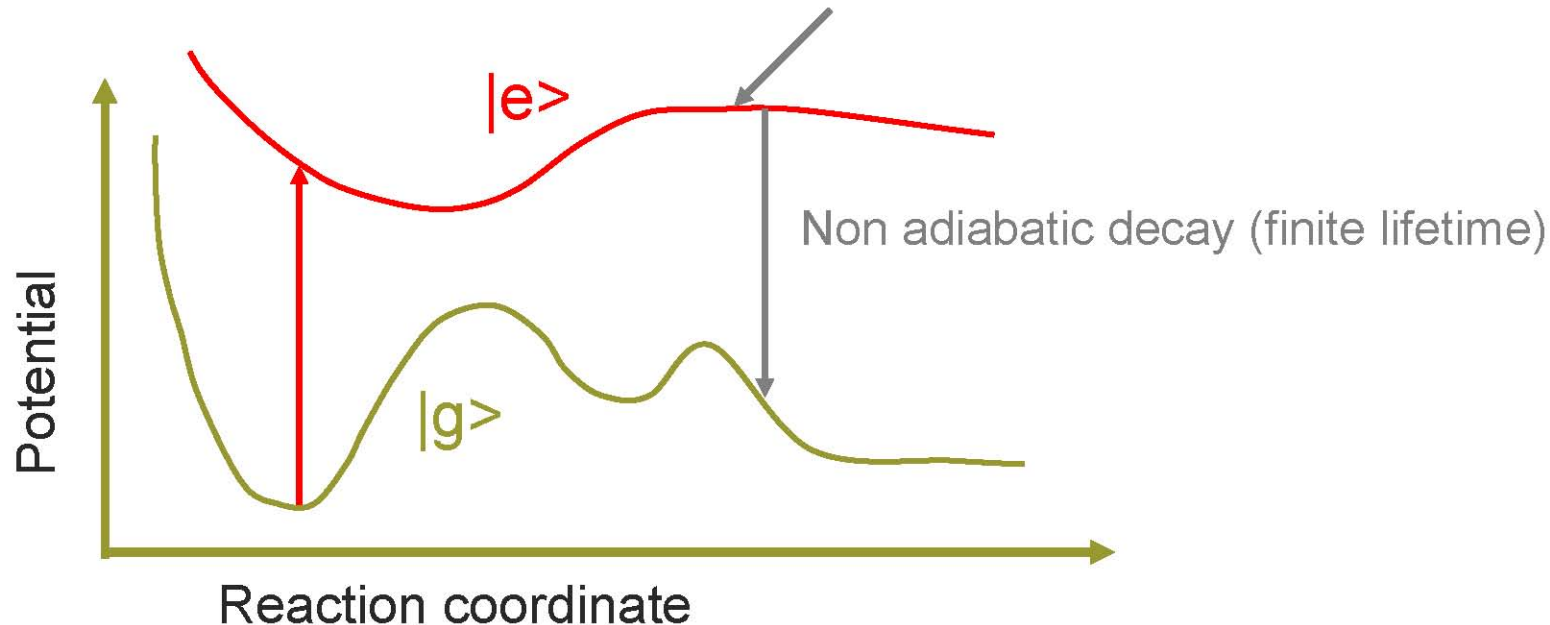
$$E_{\text{tot}} = \sum_i E_{\text{coh}}(i) = \sum_i [E_{\text{bs}}(i) + E_{\text{rep}}(i)]$$

parametrized LCAO formalism (CRT)

- **Dynamics of atoms and electrons under electronic excitations**
- Massively parallel computer architectures and suitable algorithms distribute load over processors for speed-up

System evolution on the adiabatic surface of an electronically excited state

To follow right adiabatic surfaces of excited states is rather difficult



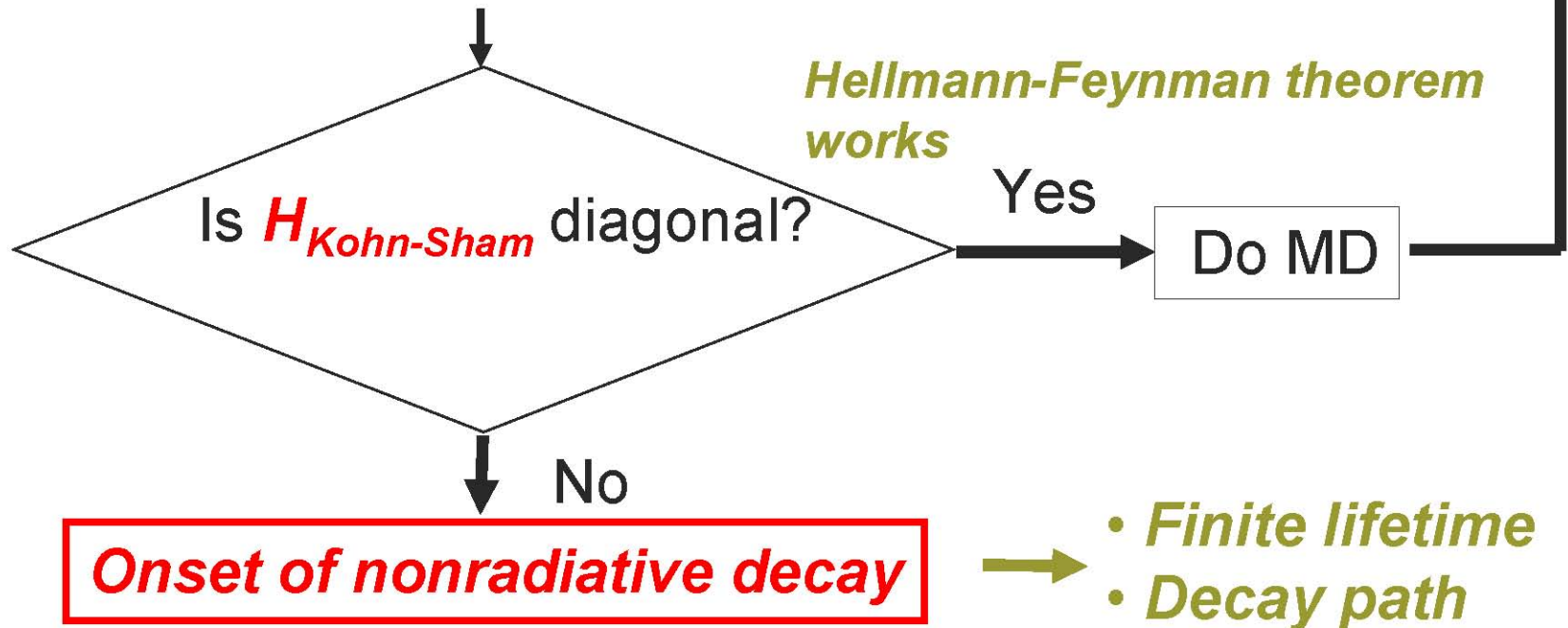
Challenges:

- ◆ Perform Molecular Dynamics simulations on the adiabatic surface of an electronically excited state
- ◆ Solve the time-dependent Schrödinger equation for electrons during ionic motion
- ◆ **F**irst-**P**inciples **S**imulation tool for **E**lectron-**I**on **D**ynamics
Sugino & Miyamoto PRB 59, 2579 (1999); PRB 66, 89901 (2002).

Excited state dynamics: flow diagram

$t = 0$: **Change level occupations** to mimic electronic excitation. Then perform static SCF calculation.

$t > 0$: Solve $\psi_n(t+\Delta t) = \exp\{-i\Delta t H(t)\} \psi_n(t)$



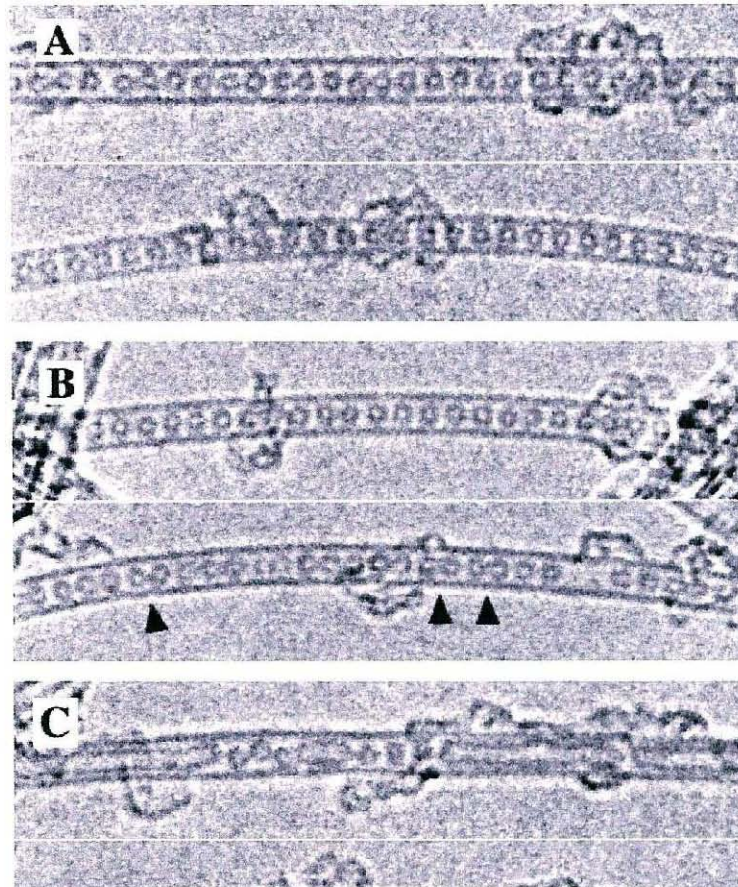
1. No need of **level assignment** for a hole and an excited electron except at the beginning.
2. **Automatic monitoring of the nonradiative decay** (lifetime, decay path) without prejudice

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Searching for Transition States in High-Dimensional Phase Spaces

Fusion of fullerenes in peapods



T=1,100°C

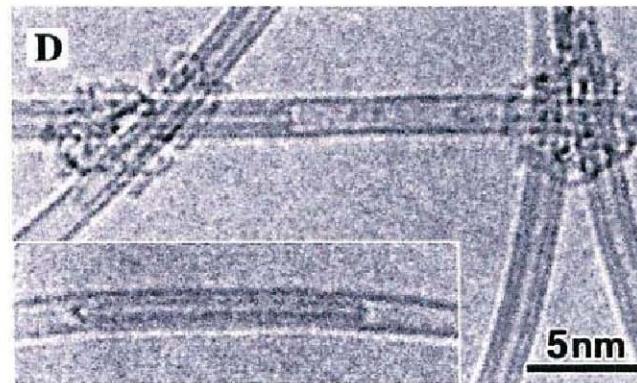
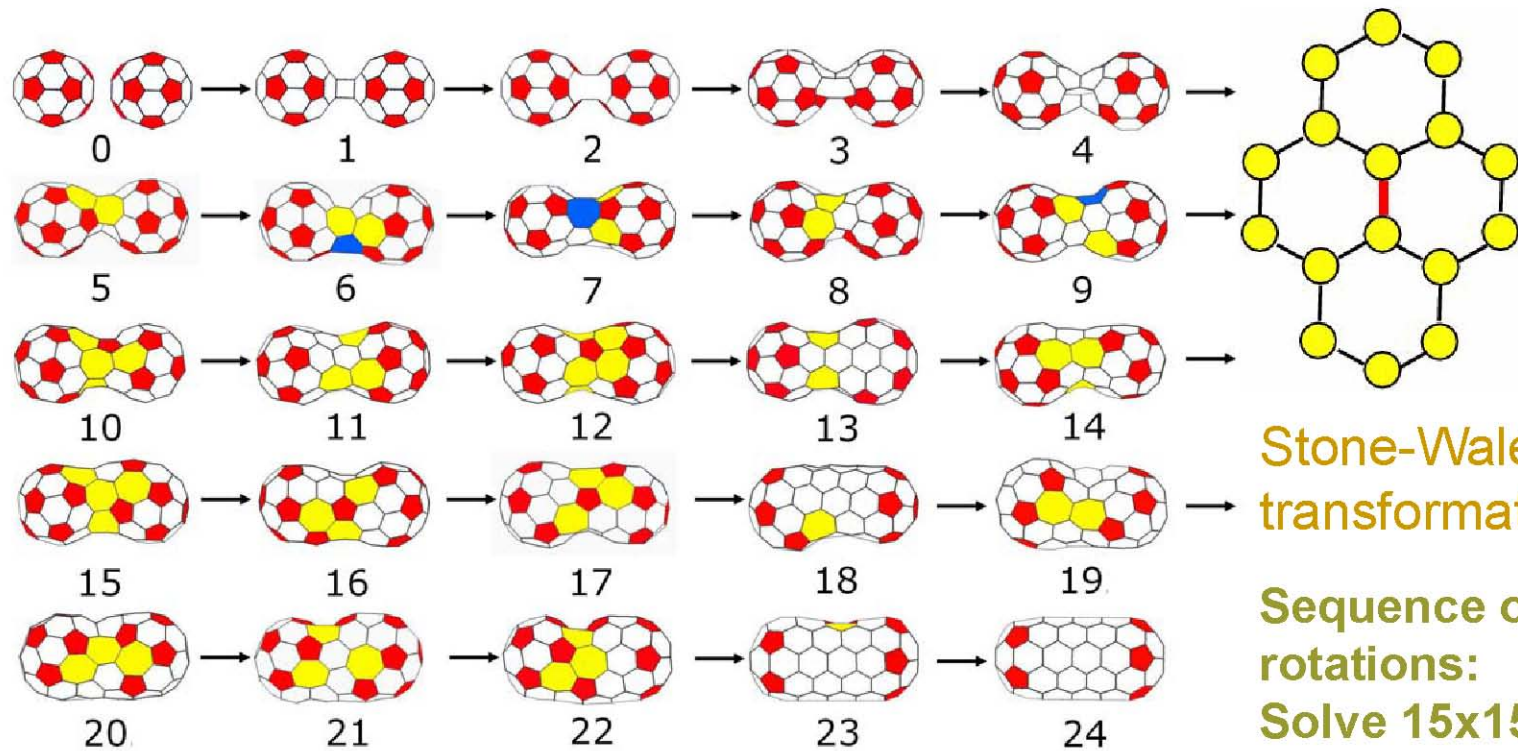


Fig. 1. Transmission electron microscopy images. (A) is for $(C_{60})_n@SWNTs$, (B) for $(C_{60})_n@SWNTs$ heated in $<10^{-6}$ Torr at 800°C for 14 h (HT800), (C) for HT100 (D) for HT1200. A and B indicate similar electron micro images, but in B we can occasionally find that some adjacent C_{60} molecules are linked together as indicate arrowheads. In C, some of the C_{60} molecules coalesce together and transform to a tubular structure. In D, no C_{60} molecule can be observed but we easily find DWNTs; in some of their inside-tubes are terminated by caps and the lengths are of order of ~ 10 nm.

[S. Bandow, M. Takizawa, K. Hirahara, M. Yudasaka, and S. Iijima, Chem. Phys. Lett. 337, 48 (2001)]

Stone-Wales rearrangement pathway for fusion of fullerenes

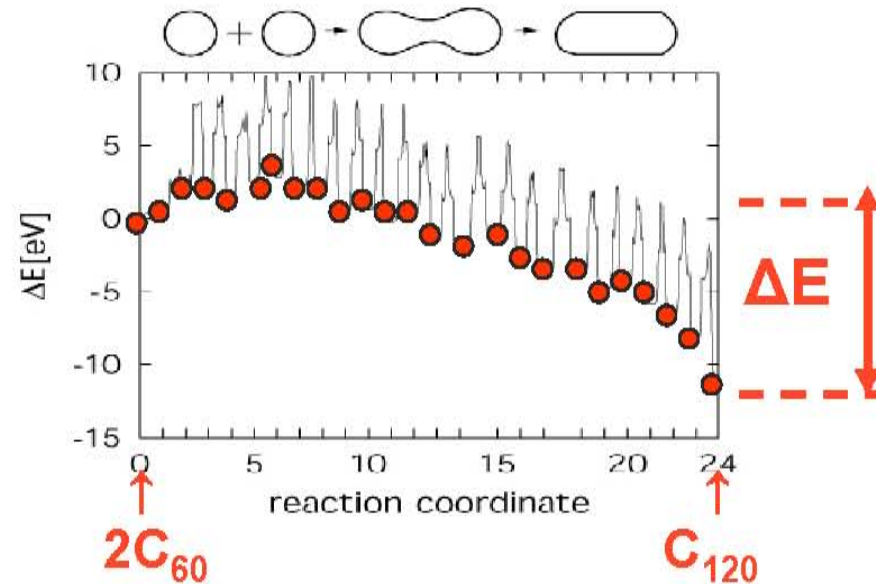
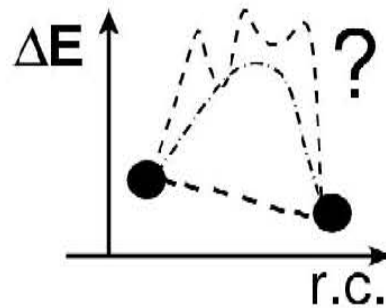
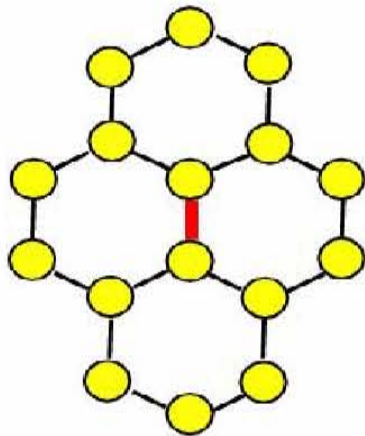
[Hiroshi Ueno, Shuichi Osawa, Eiji Osawa, and Kazuo Takeuchi, *Fullerene Science and Technology* **6**, 319-338 (1998)]



Sequence of bond rotations:
Solve 15x15x15
Rubik's Cube puzzle

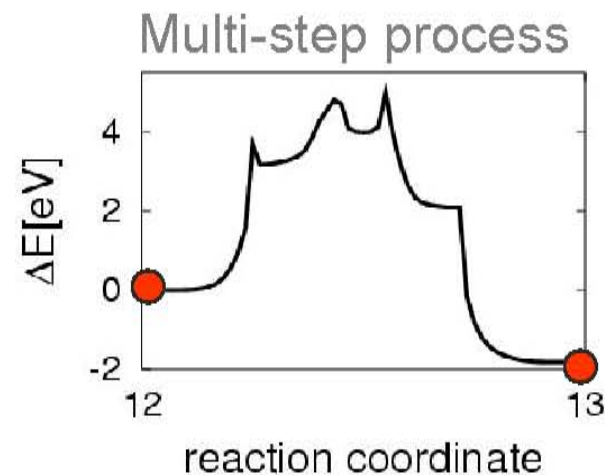
❖ Do we understand the energetics?

Do we understand the Stone-Wales process?

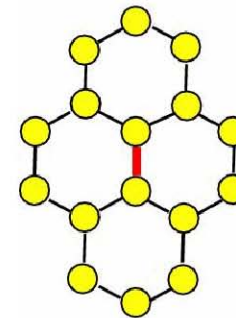
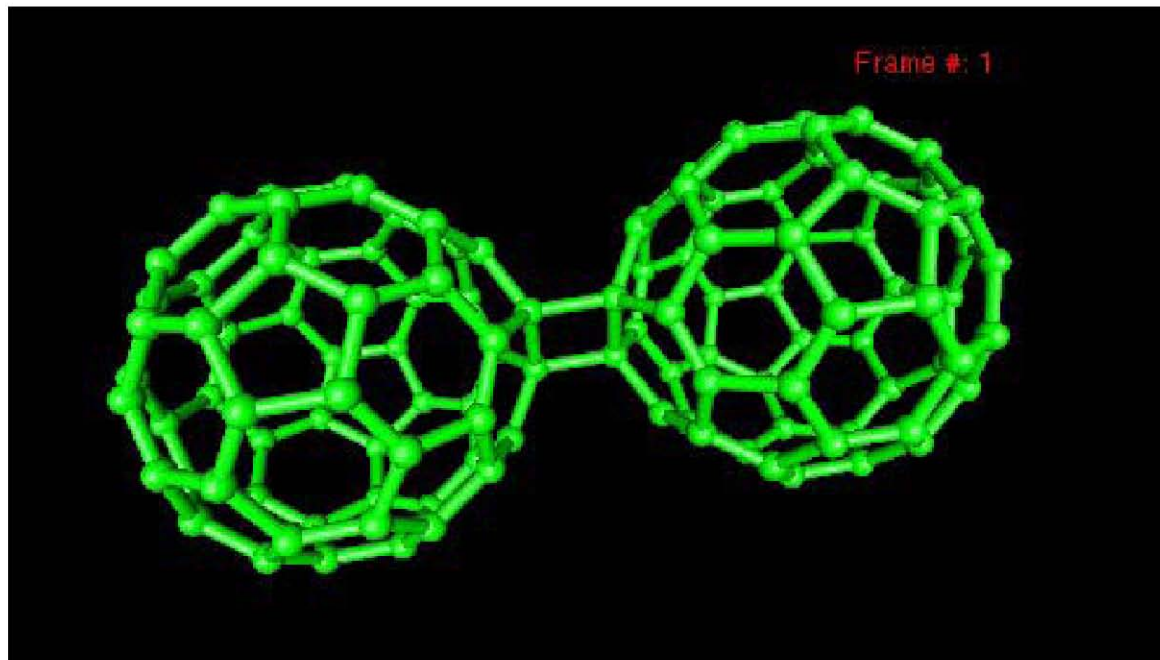


Search in 360-dimensional configuration space using string method:

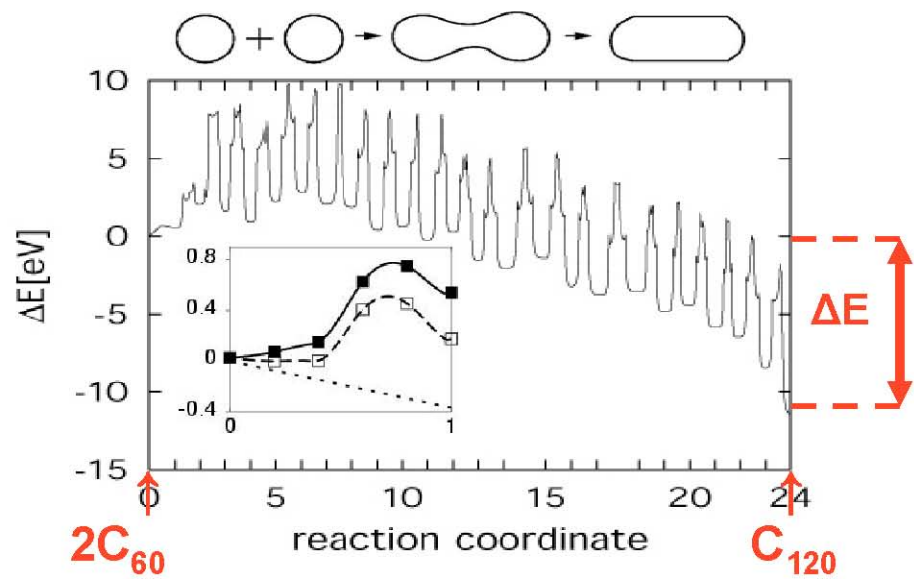
- Stone-Wales is a multi-step process
- Activation barriers do not exceed ≈ 5 eV



Minimum energy path for the $2C_{60} \rightarrow C_{120}$ fusion



Sequence of
Stone-Wales
transformations



❖ Conclusions:

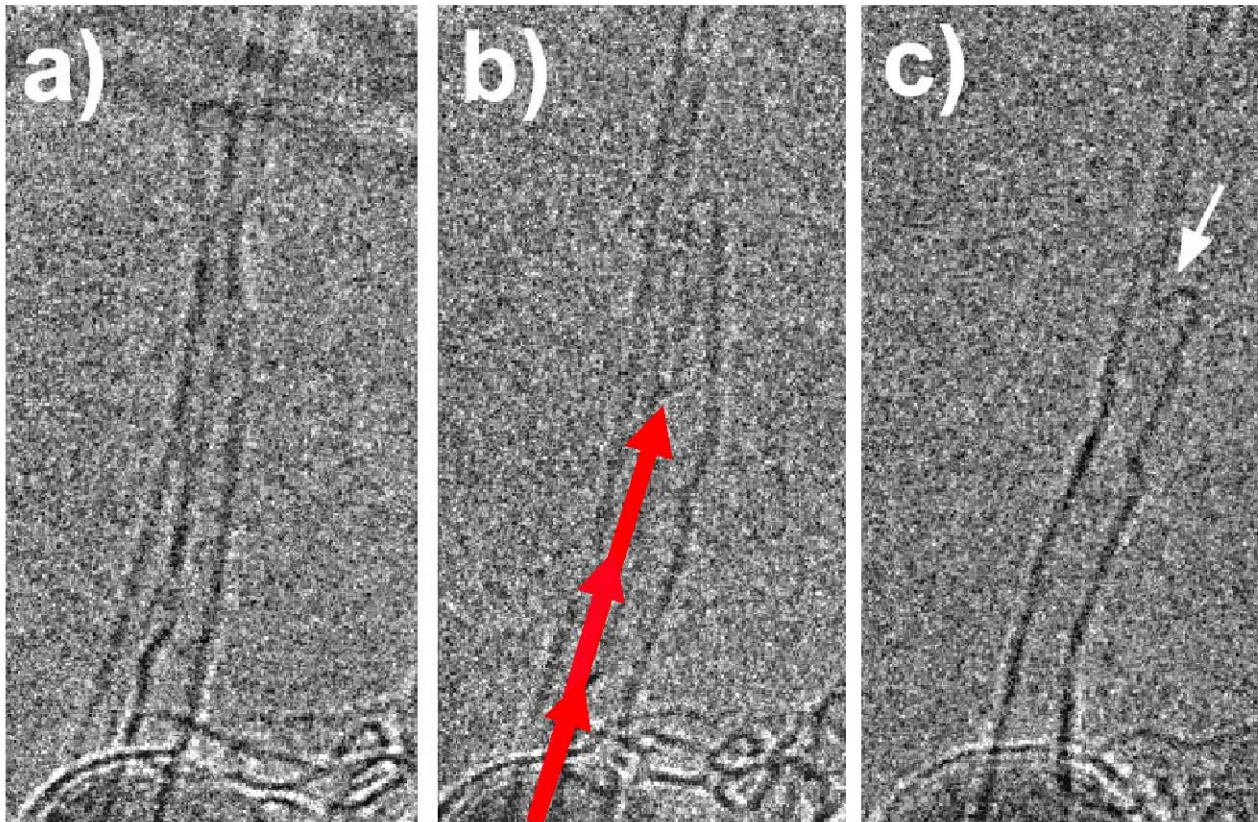
- Fusion is exothermic. Energy gain $\Delta E \approx 1\text{ Ry}$.
- Essential initial step: (2+2) cycloaddition

Seungwu Han, Mina Yoon, Savas Berber, Noah Park, Eiji Osawa, Jisoon Ihm, and David Tománek, Microscopic Mechanism of Fullerene Fusion, Phys. Rev. B **70**, 113402 (2004).

Fusion of nanotubes

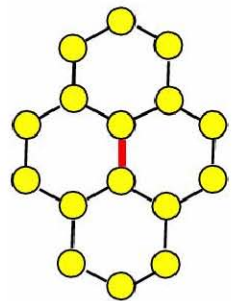
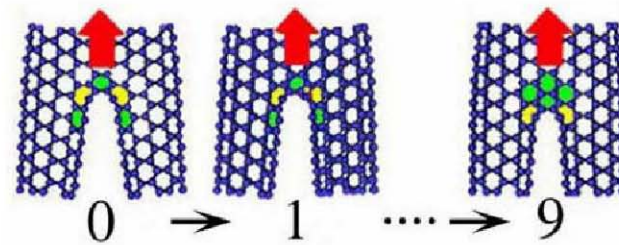
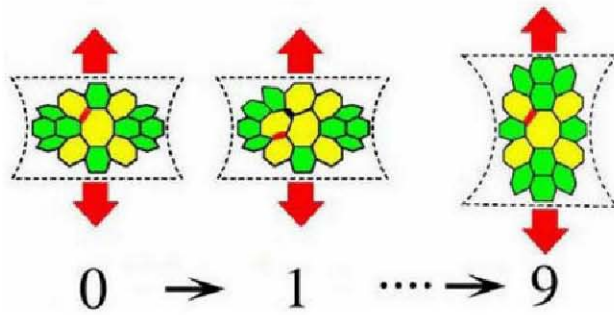
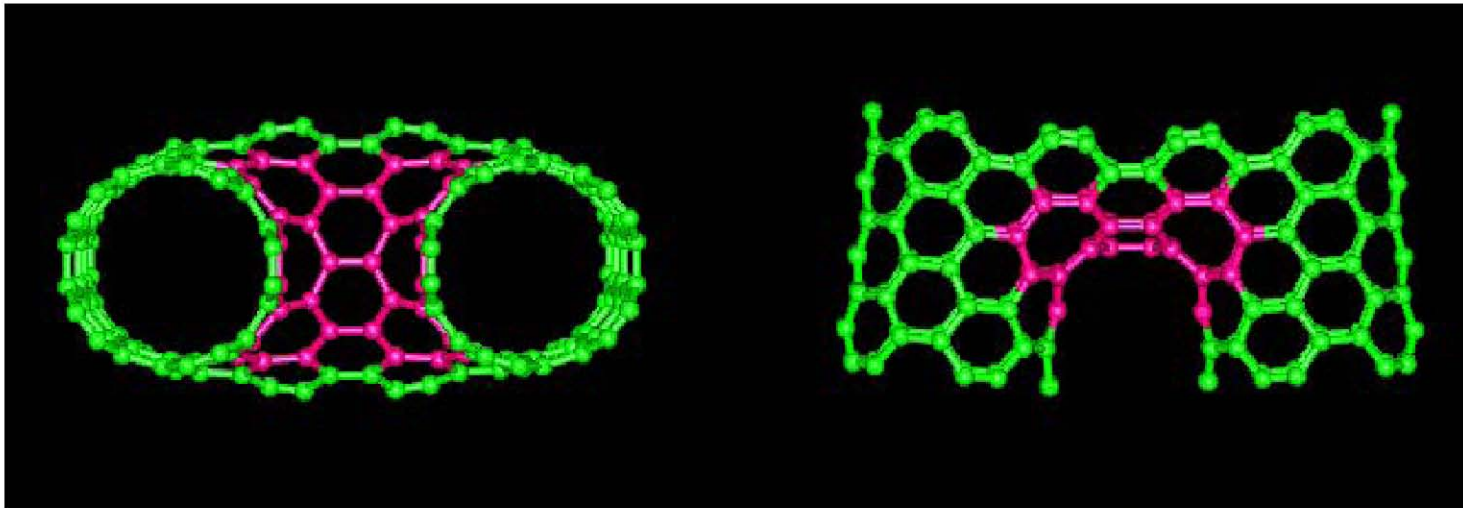
The zipper mechanism

M. Yoon, S. Han, G. Kim, S. Lee, S. Berber, E. Osawa, J. Ihm, M. Terrones, F. Banhart, J.-C. Charlier, N. Grobert, H. Terrones, P. M. Ajayan, D. Tománek, Phys. Rev. Lett. **92**, 075504 (2004).

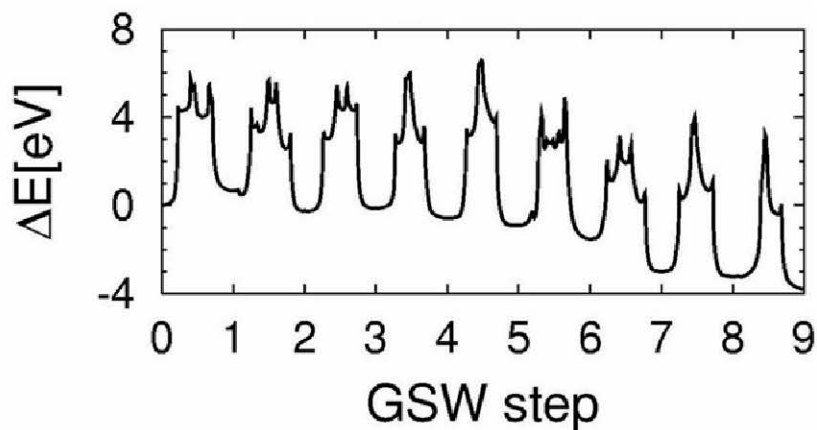


Zipper

Minimum energy path for the $(5,5)+(5,5) \rightarrow (10,10)$ fusion



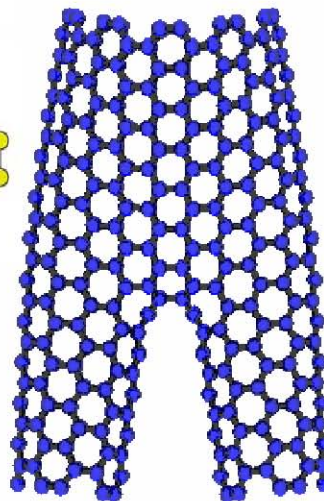
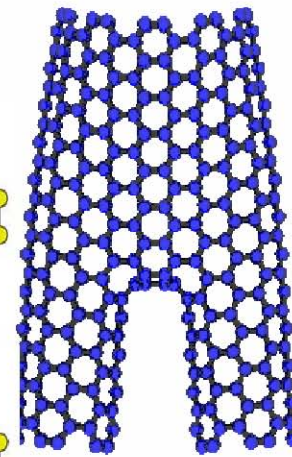
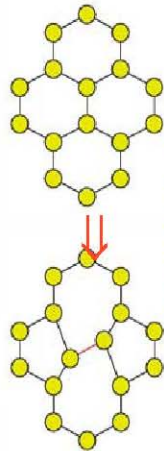
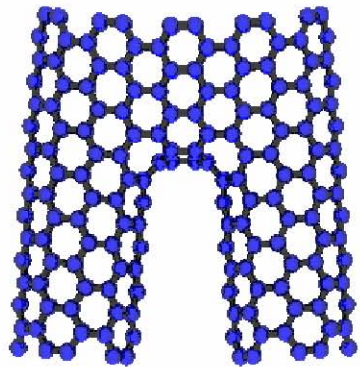
Sequence of
Stone-Wales
transformations



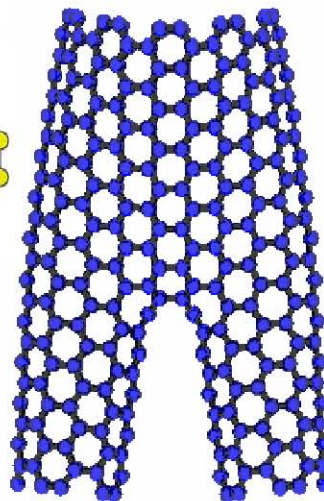
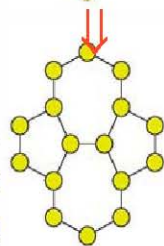
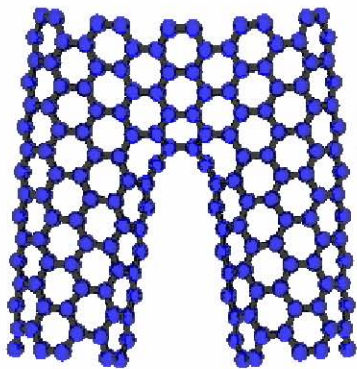
❖ Conclusion: Fusion is exothermic

Geometry of fusing Nanopants

Type A:
6 heptagons
in junction
area

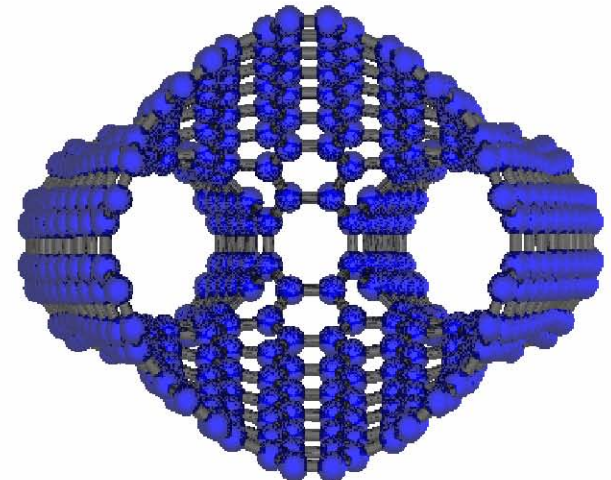
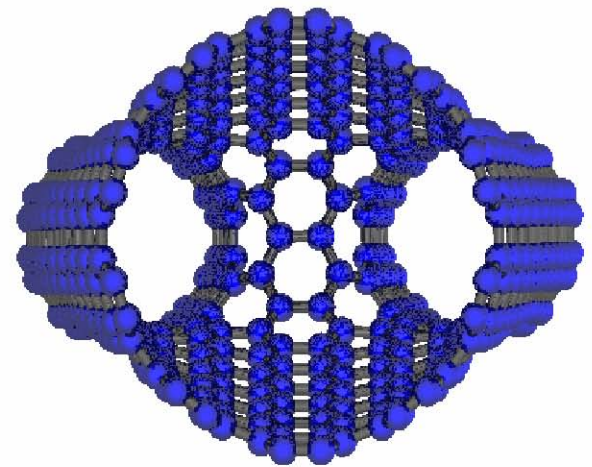


Type B:
1 octagon,
4 heptagons
in junction
area



front view

top view

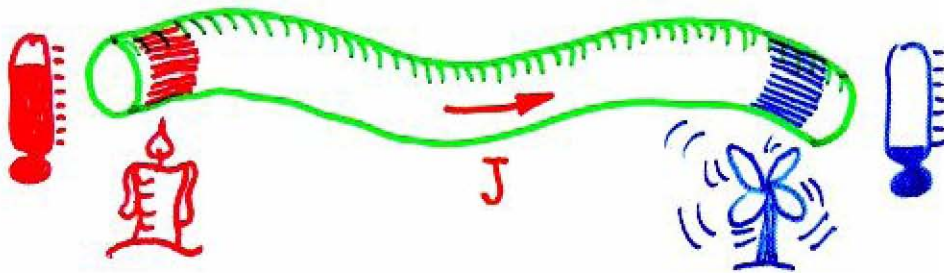


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David Tománek, Carbon-based nanotechnology on a supercomputer, Topical Review in J. Phys.: Condens. Matter 17, R413-R459 (2005).

Ground State Molecular Dynamics Simulations

Thermal Conductivity of Carbon Nanotubes



Savas Berber, Young-Kyun Kwon, and David Tománek,
Phys. Rev. Lett. 84, 4613
(2000)

- ◆ Nanotubes may help solve the heat problem:

Efficient conductors of electrons and heat

- ◆ Record Heat Conductivity:

- * Diamond

(isotopically pure): 3320 W/m/K

- * Nanotubes: 6,600 W/m/K (theory, SWNT)

>3,000 W/m/K (experiment, MWNT)

(room temperature values)

(combination of large phonon mean free path,
speed of sound, hard optical phonon modes)

Direct molecular dynamics simulation

$$J = -\kappa S \frac{\partial T}{\partial x} \quad \text{Fourier's law}$$

J: Heat flux

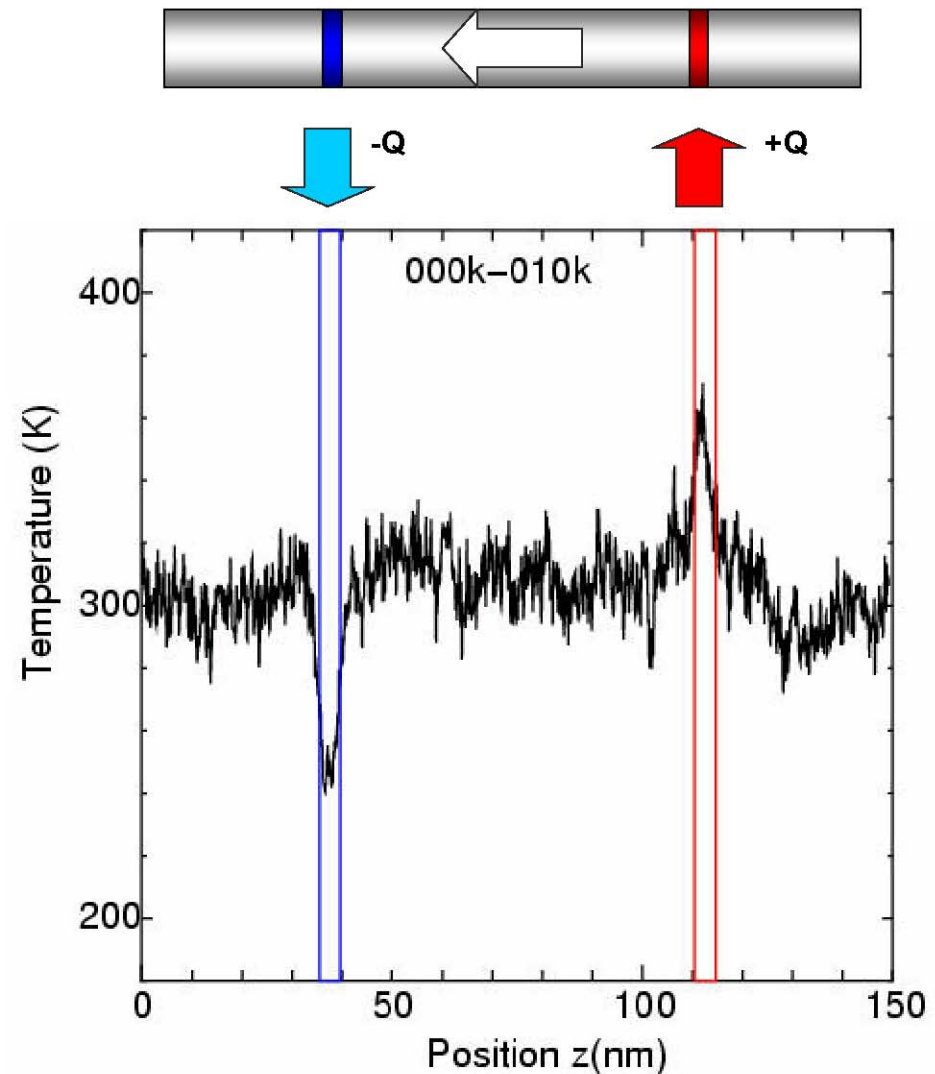
S: Cross section area

κ : Thermal conductivity

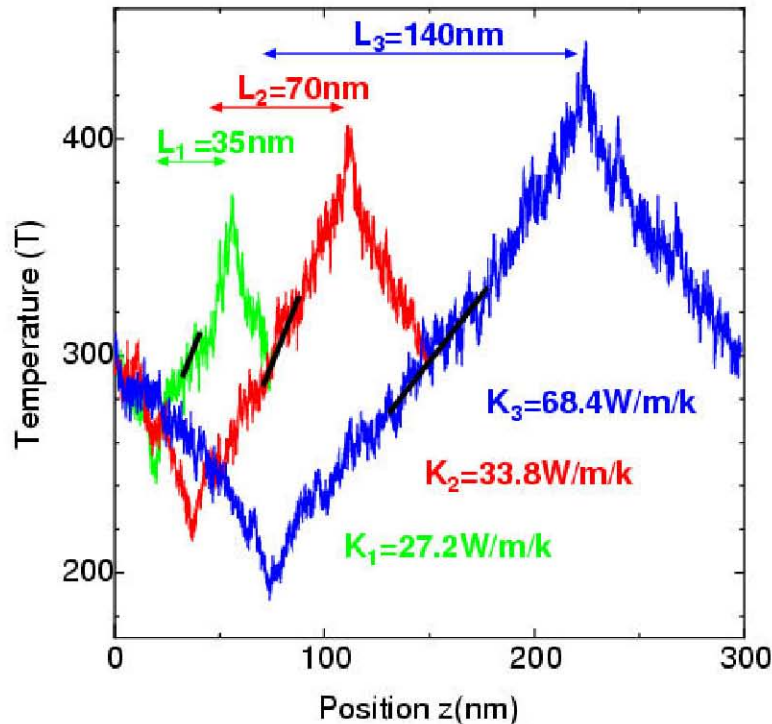
$\kappa \propto \text{phonon mean free path } \lambda$
(distance that phonons travel without scattering)

■ Problem: Hot-cold spot separation $> \lambda$ to avoid artifacts

■ How large is λ ?



CPU resources for a large-scale simulation



One time step in MD
 $\propto N$ (number of atoms)

Thermal flux propagation time
along unit cell \propto (unit cell length) $\propto N$

Total CPU time $\propto N^2$

CPU Time for L_1 :	2.5 days	Simulation time: 60ps	$N \approx 25,000$
L_2 :	10 days	120ps	$N \approx 50,000$
L_3 :	40 days	240ps	$N \approx 100,000$

(Parametrized LCAO MD calculations)

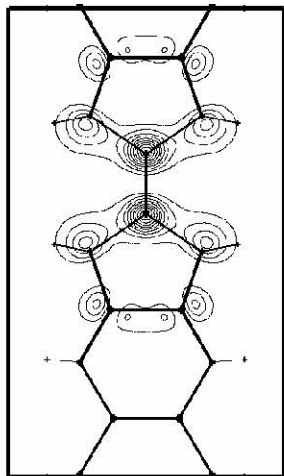
1024 Processors / 2.5 Teraflops

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Excited State Molecular Dynamics Simulations

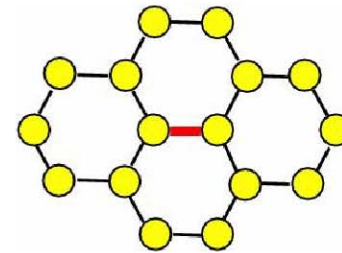
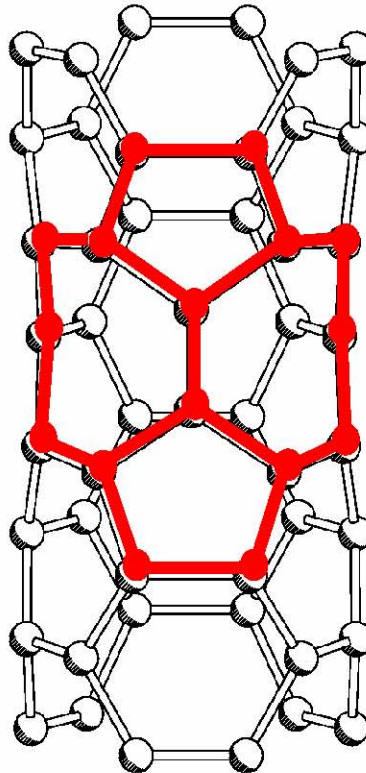
Detection of Stone-Wales defects



π^* state
(electron)

6 eV

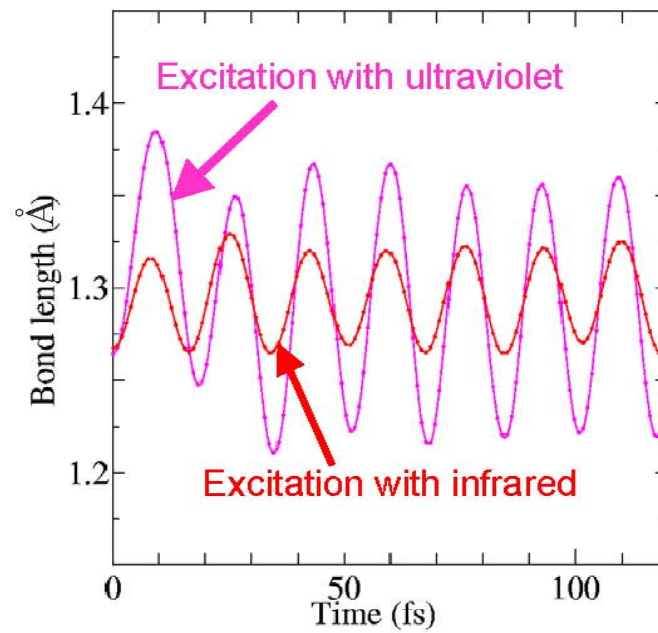
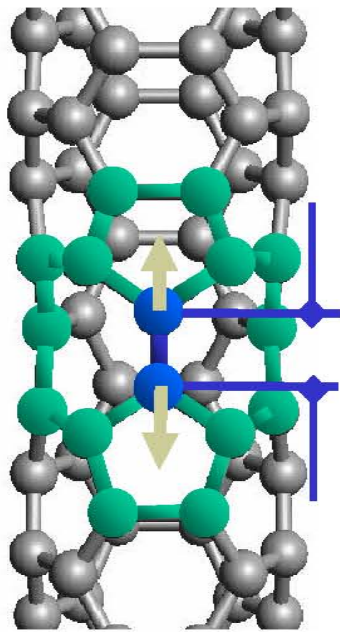
σ - π
hybridize
d state
(hole)



Origin of
Stone-Wales
defect

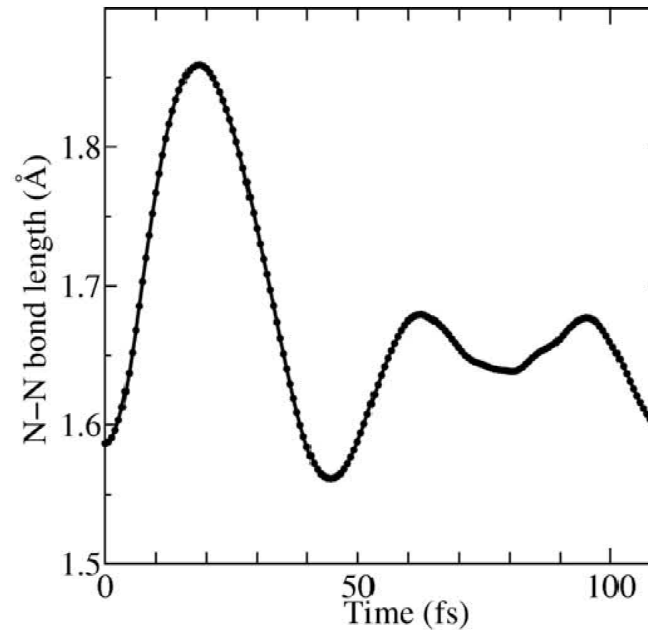
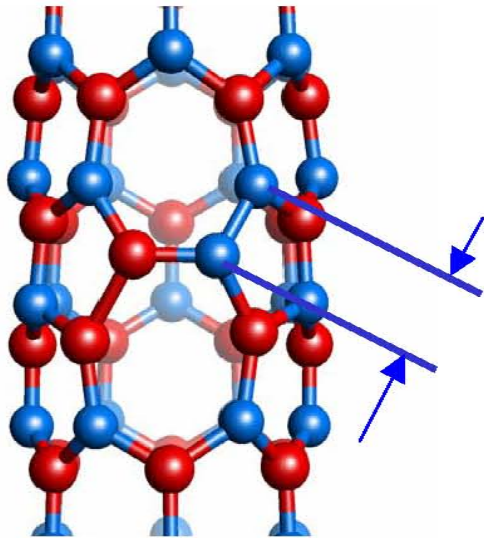
► How does a Stone-Wales defect react under photo-excitations?

(3,3)
carbon
nanotube



T=17 fs
(1962 cm⁻¹)

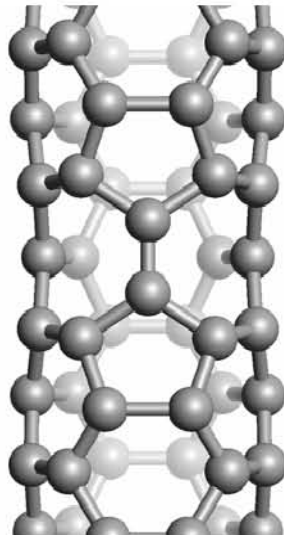
(3,3) BN
nanotube



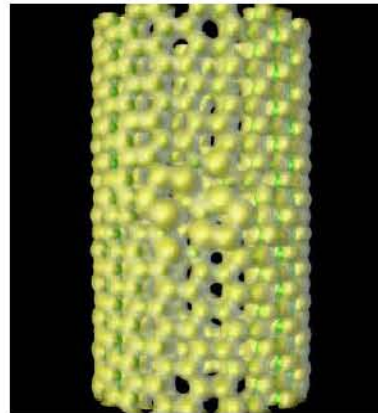
Stone-Wales defects are not removed, but can be identified using photo-excitations

STM characterization of Stone-Wales defects

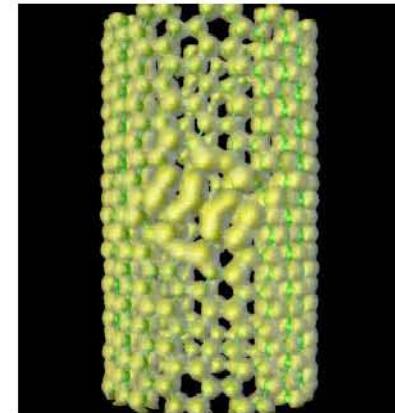
(3,3) carbon
nanotube



$V = -1.5 \text{ V}$

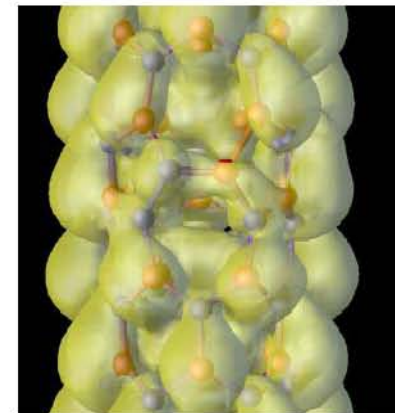
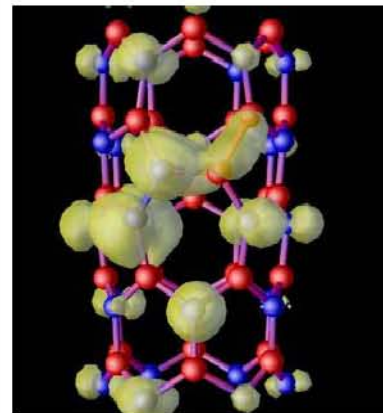
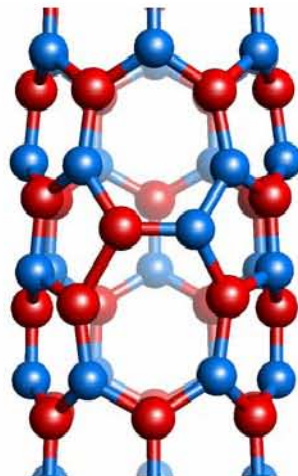


$V = +1.5 \text{ V}$



(10,10)
CNT

(3,3) BN
nanotube

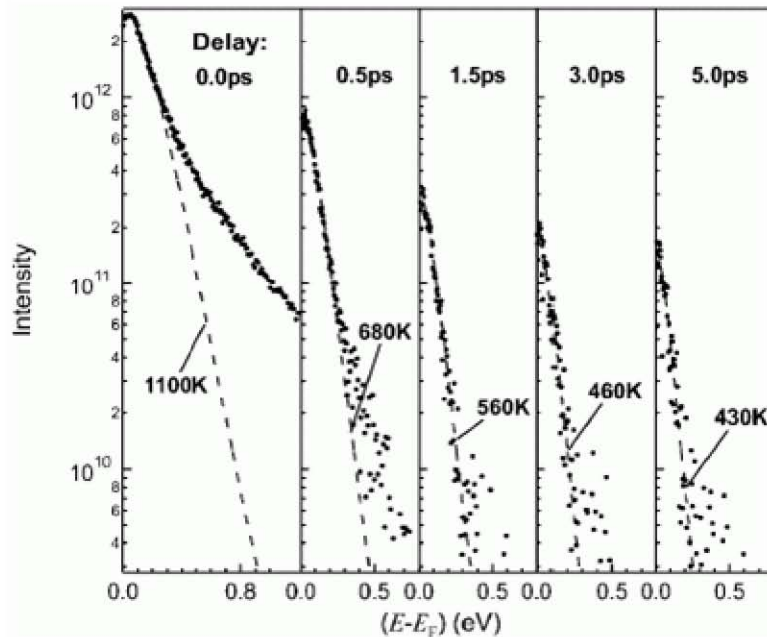
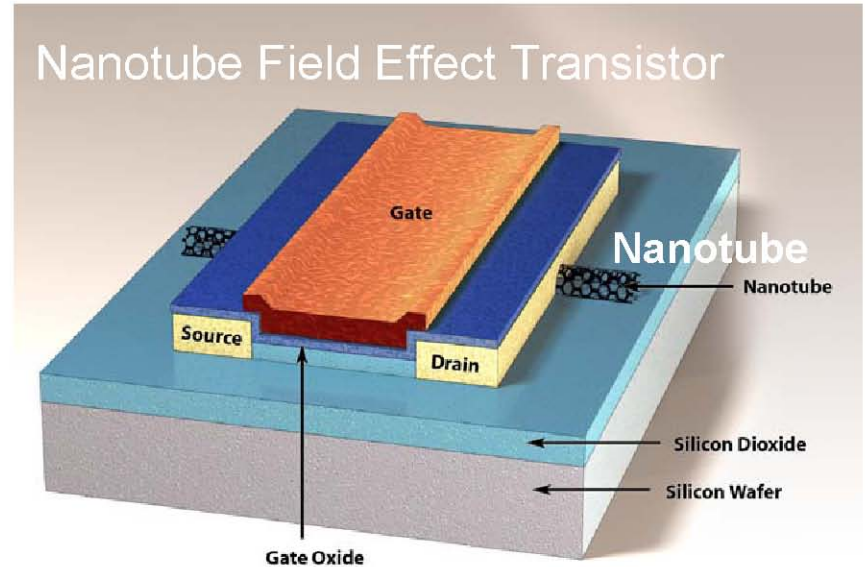


(3,3)
BNNT

Y. Miyamoto, A. Rubio, S. Berber, M. Yoon, and D. Tománek,
Phys. Rev. B 69, 121413 (2004).

Hot carrier dynamics in nanotubes

- How useful are carbon nanotube devices (field-effect transistors, non-linear optical devices)?
- Maximum switching frequency:
 - ➔ **lifetime of excited carriers**
- How long do electronic excitations last?
- What dampens electronic excitations:
 - Electron gas?
 - Phonons?



Evolution of photoelectron spectra as a function of pump-probe delay. At pump-probe delays of over 200 fs, the spectra can be well described by a Fermi-Dirac distribution (dashed lines).

T. Hertel and G. Moos,
PRL **84**, 5002 (2000)

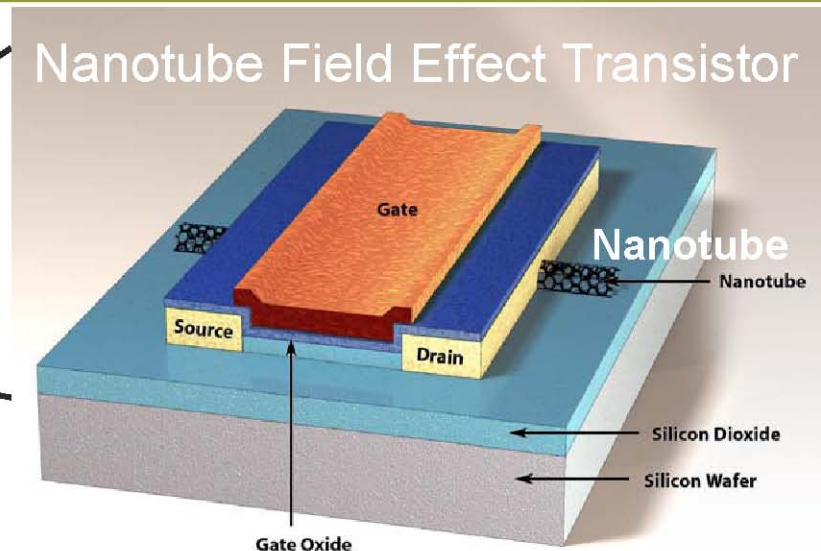
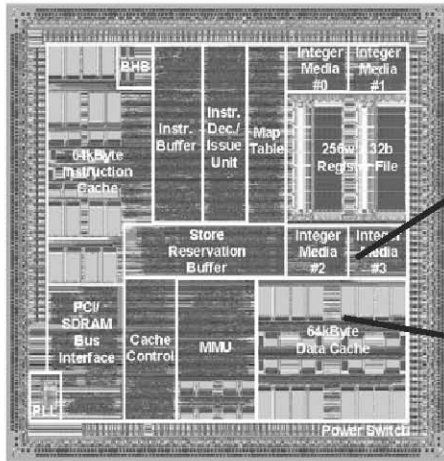
Interpretation:
e-e comes before e-ph

Outline

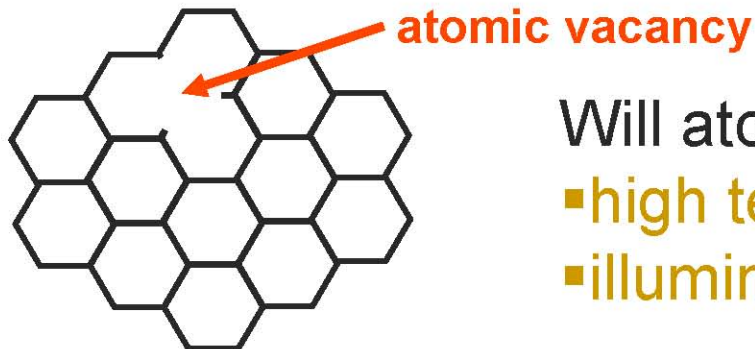
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What are Excitations Good for?

Defect tolerance of nanotubes



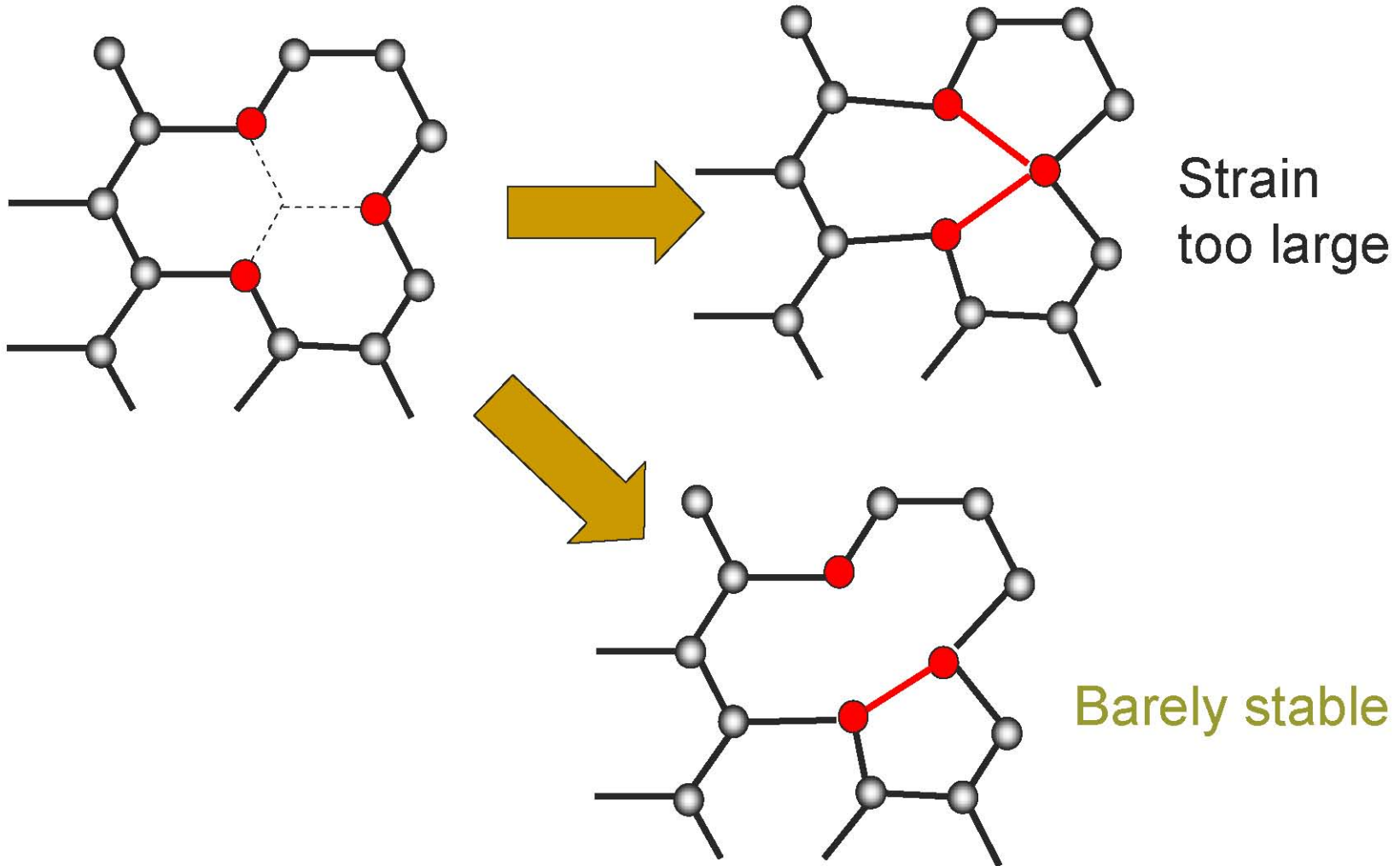
- Defects limit performance, lifetime of devices
- Are CNT devices as sensitive to defects as Si-LSI circuits?



Will atomic vacancies trigger failure under

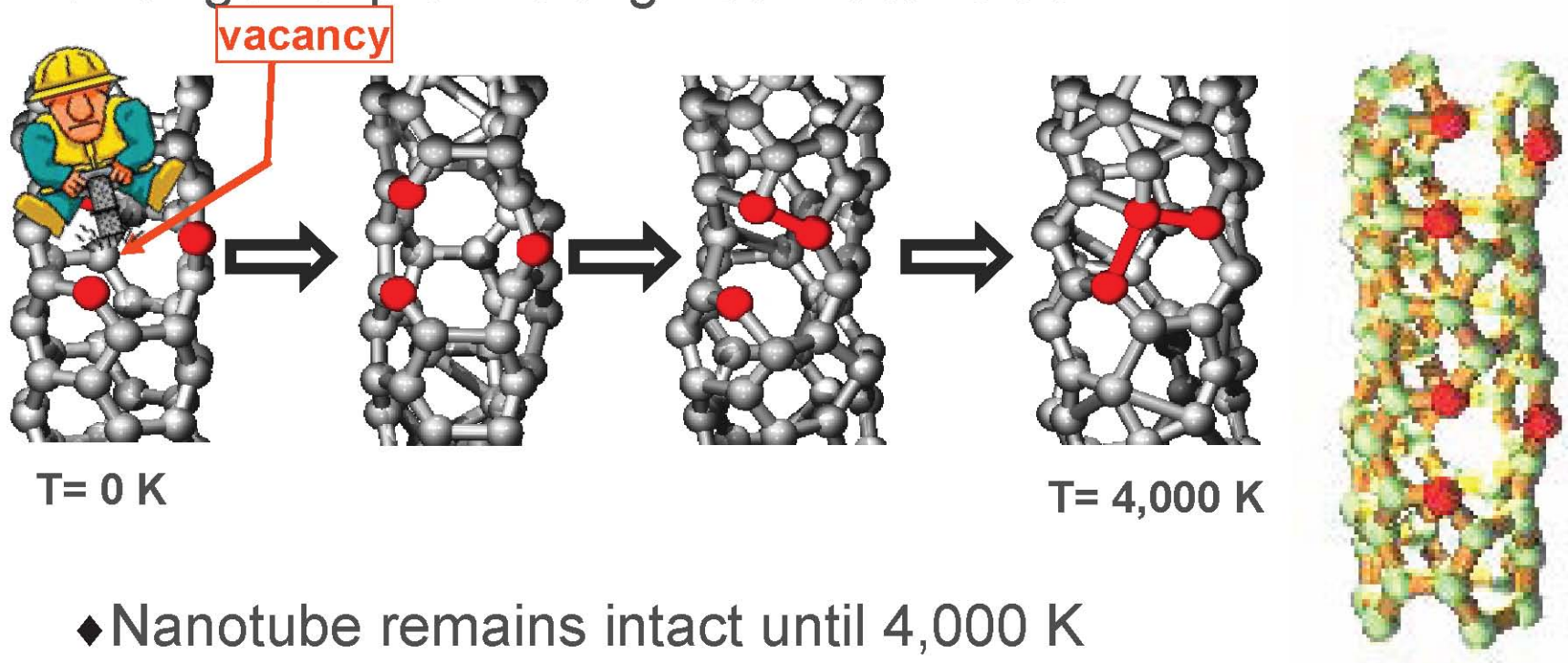
- high temperatures?
- illumination?

Equilibrium structure near a monovacancy in sp^2 carbon



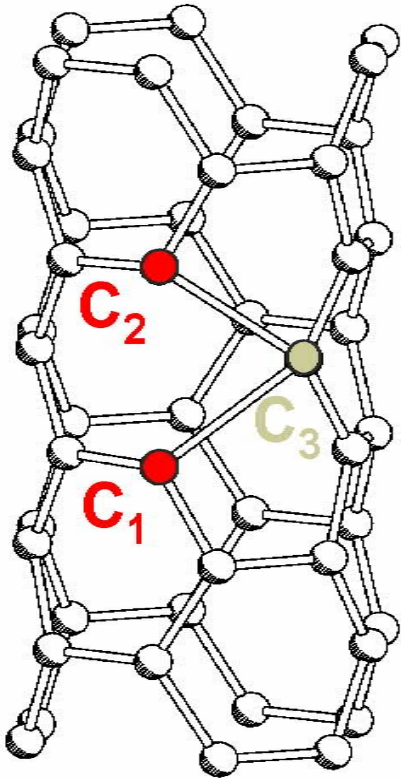
Stability of defective tubes at high temperatures

- ◆ Danger of pre-melting near vacancies?



- ◆ Nanotube remains intact until 4,000 K
- ◆ **Self-healing** behavior:
Formation of new bond helps recover
 - structural stiffness
 - conductance

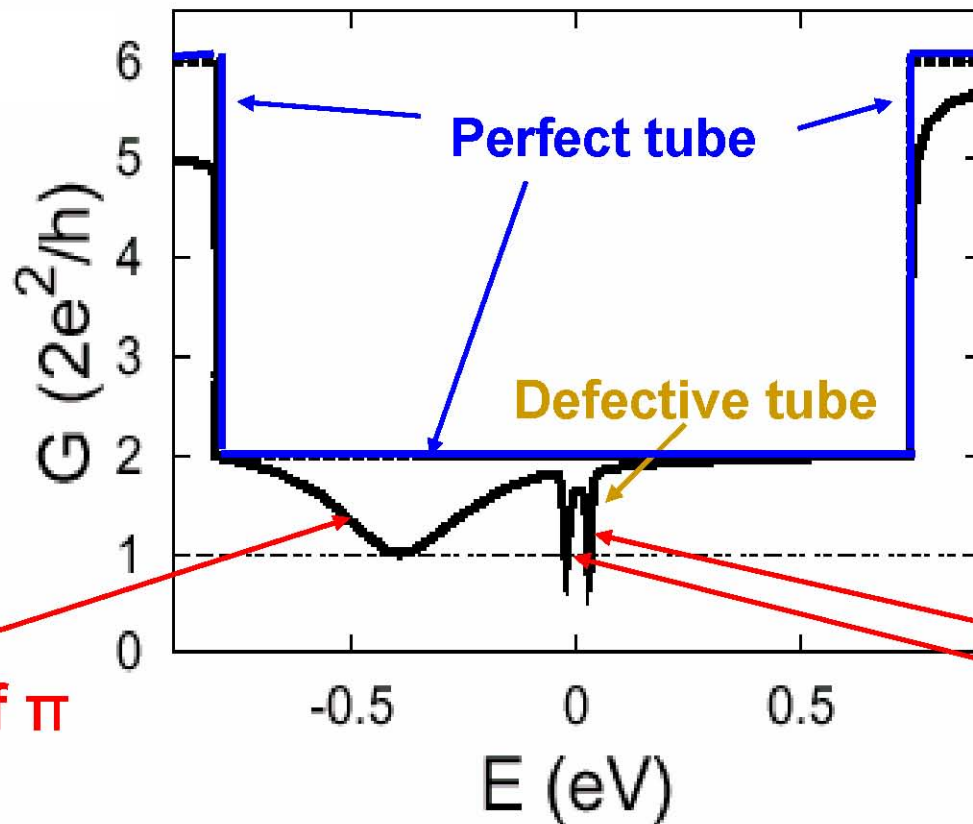
Reconstructed geometry



Stability increase due to reconstruction
(bond formation across vacancy)

Does reconstruction affect favorably transport in defective tubes?

Quantum conductance of a (10,10) nanotube with a single vacancy



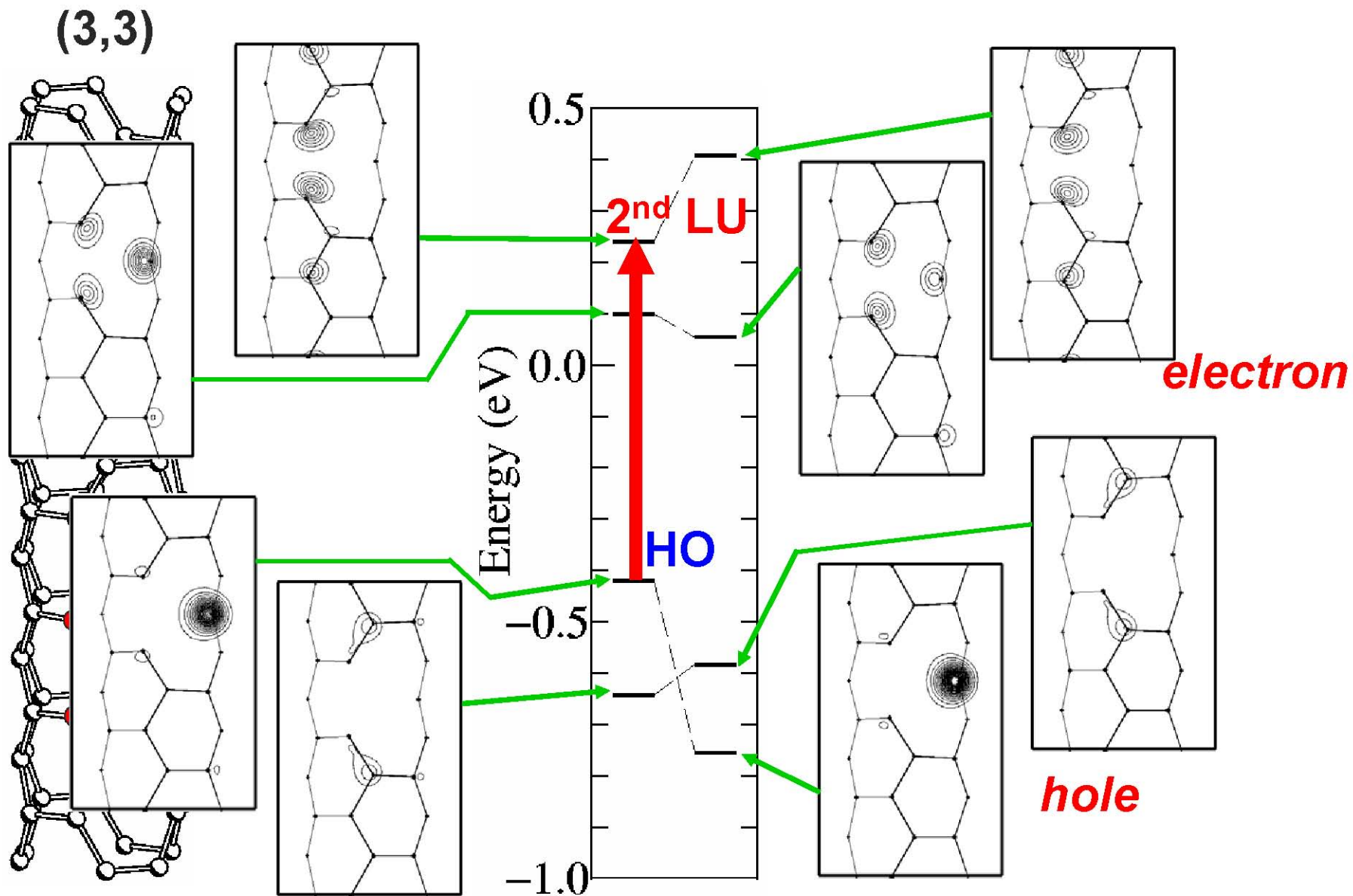
Choi, Ihm,
Louie, Cohen,
PRL (2000)

**Missing
network of π
electrons**

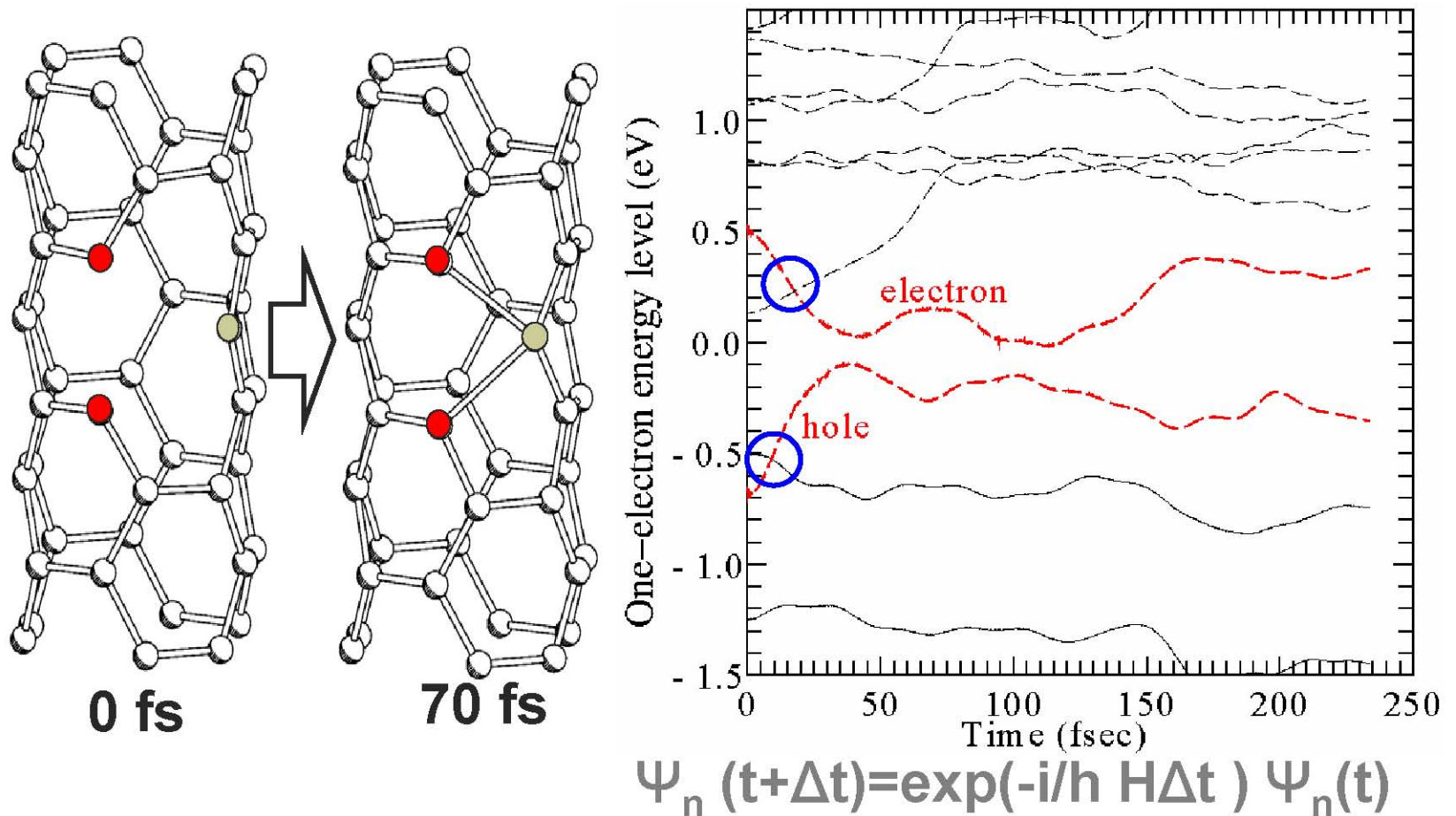
**Dangling
bonds: σ
electrons**

Good news for applications: Self-healing
by reconstruction may remove one of the sharp dips

Optical excitation ($\Delta E = 0.9$ eV)

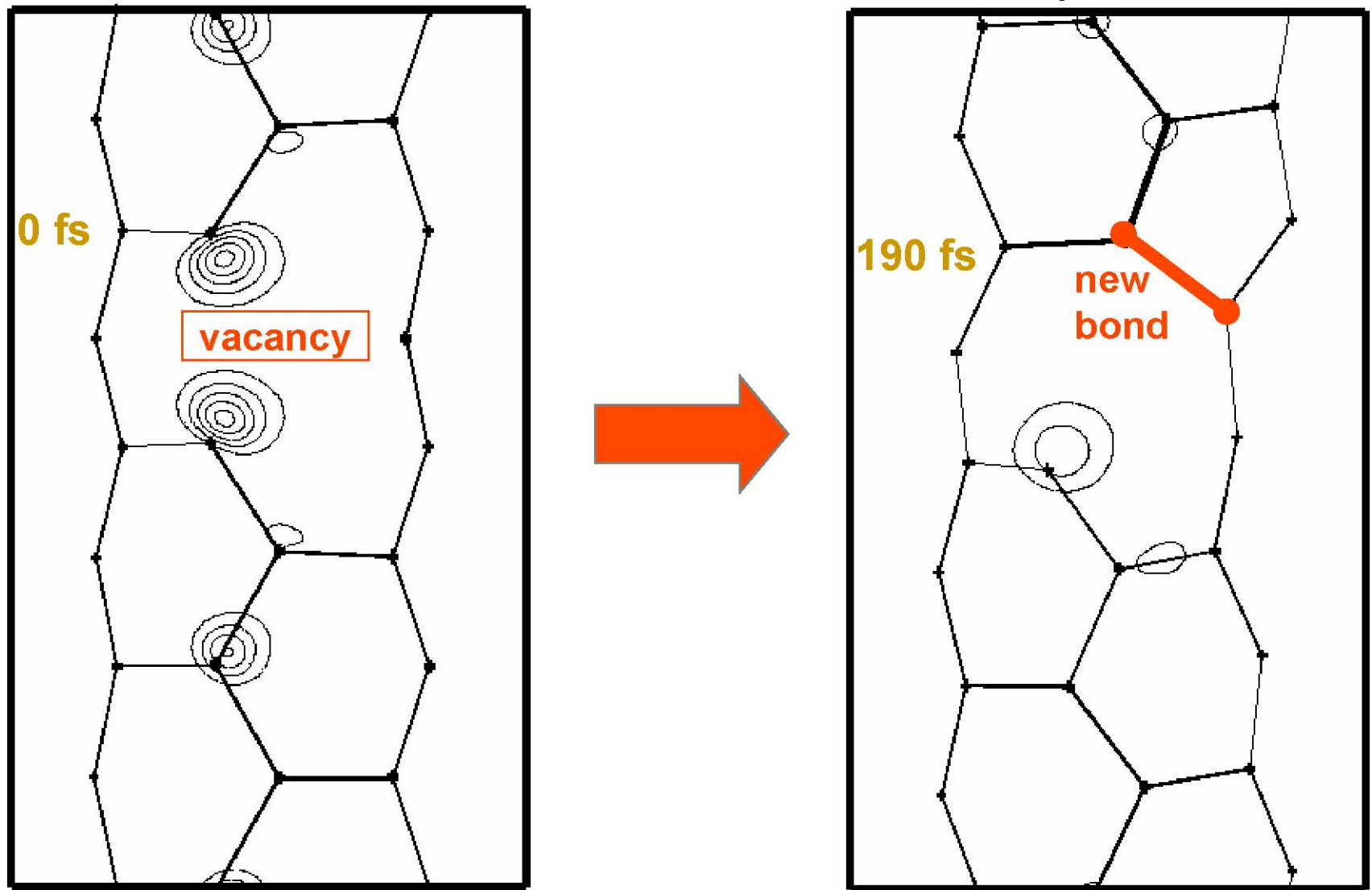


Time evolution of the electronic states



- ◆ Very long-lived excitation
- ◆ Correct PES is followed in case of level alternation

Structural changes under illumination

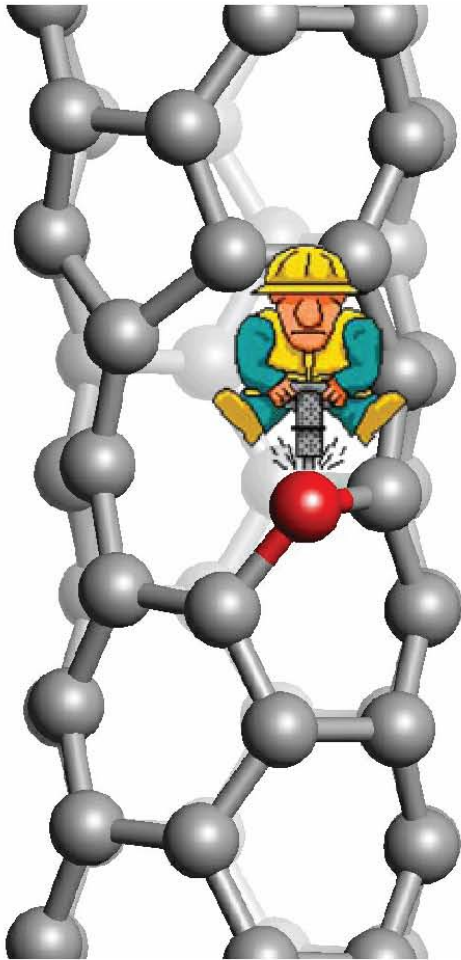


◆ Self-healing due to new bond formation

Y. Miyamoto, S. Berber, M. Yoon, A. Rubio, D. Tománek, Can Photo Excitations Heal Defects in Carbon Nanotubes? Chem. Phys. Lett. 392, 209–213 (2004)

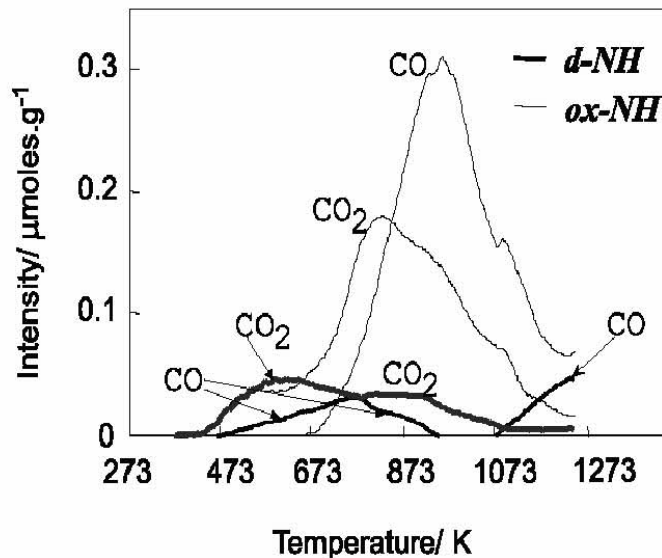
Deoxidation of defective nanotubes

How to deoxidize?

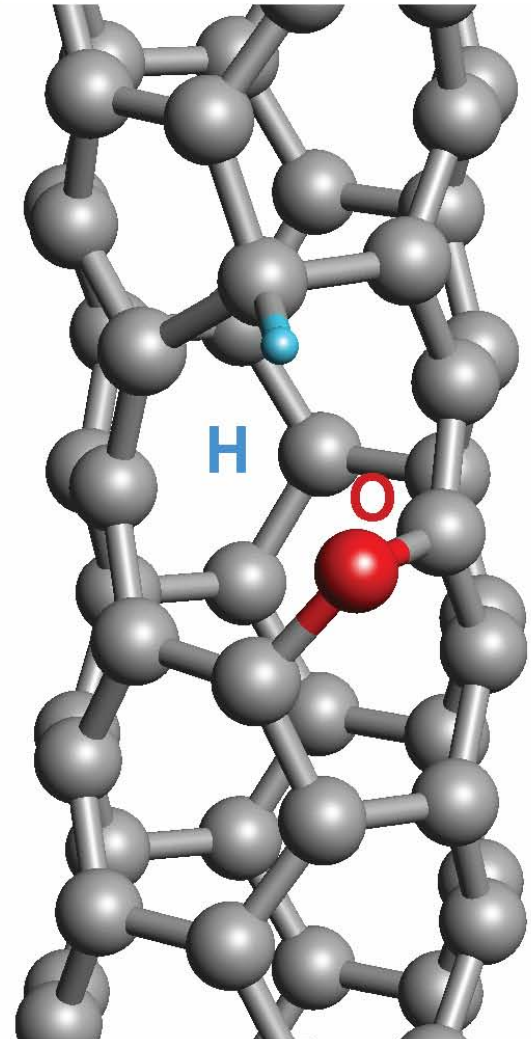


▪By heat treatment?

⇒No: Larger damage to nanotube



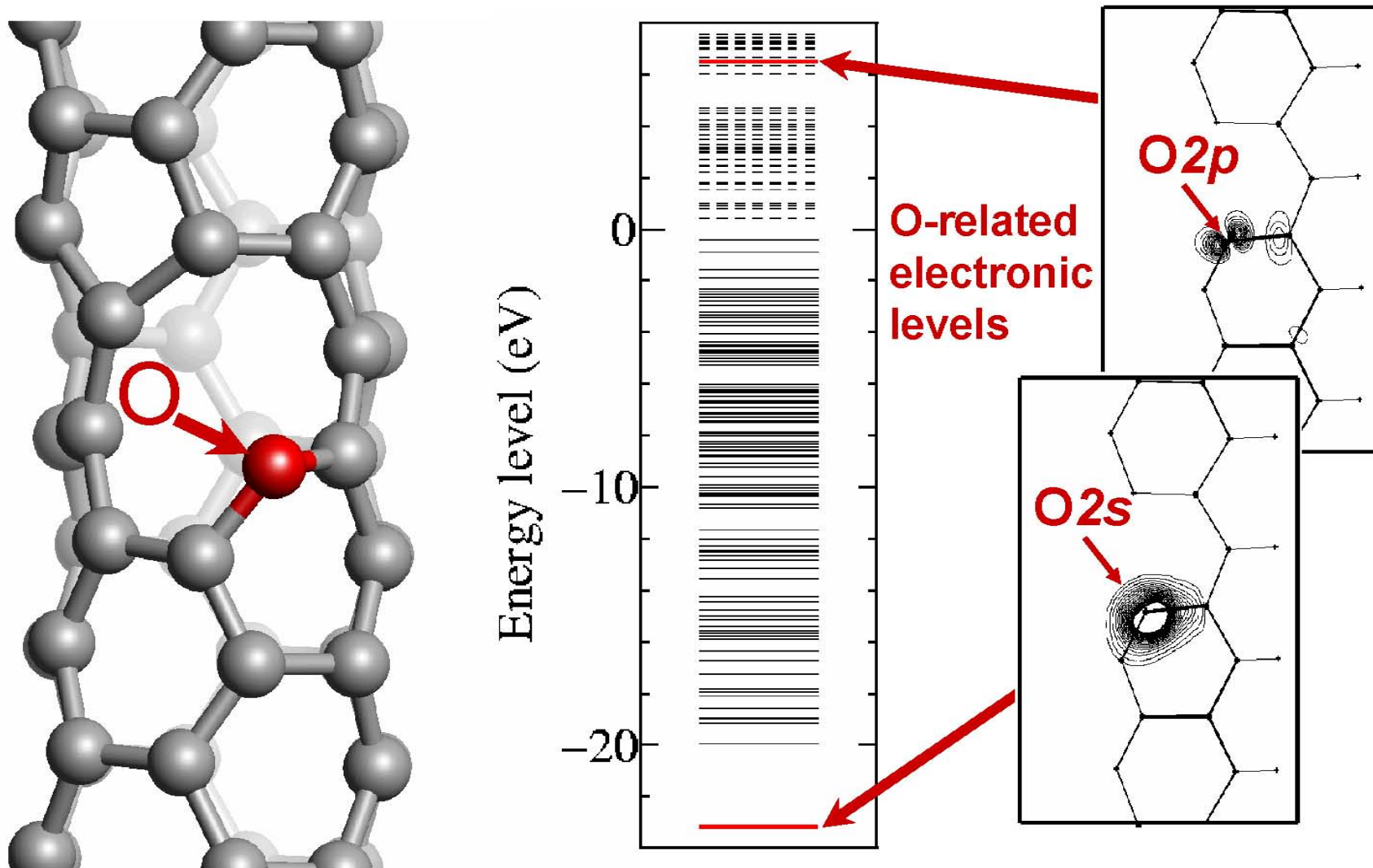
▪By chemical treatment with H?



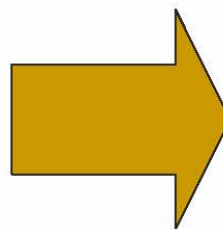
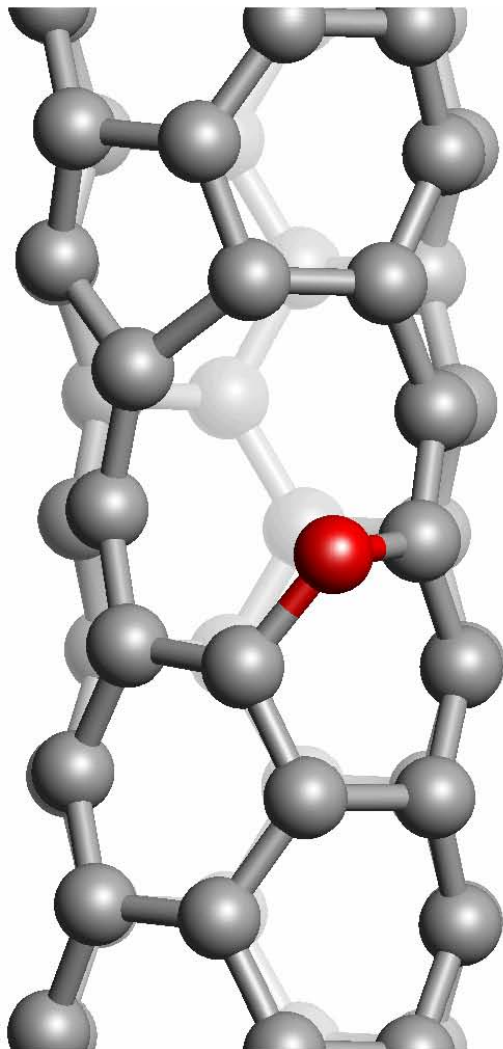
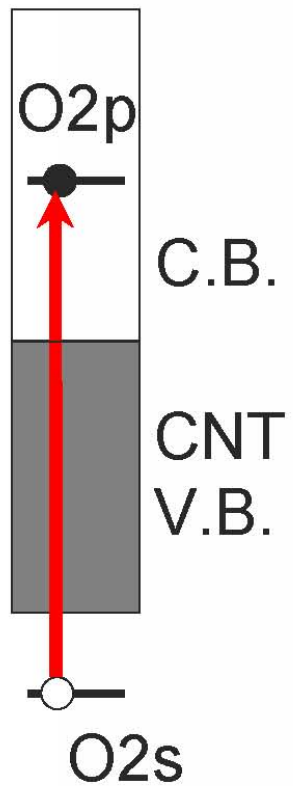
Yoshiyuki Miyamoto, Noboru Jinbo,
Hisashi Nakamura, Angel Rubio, and David Tománek,
Photosurgical Deoxidation of Nanotubes, Phys. Rev. B 70 (2004).

Alternative to thermal and chemical treatment

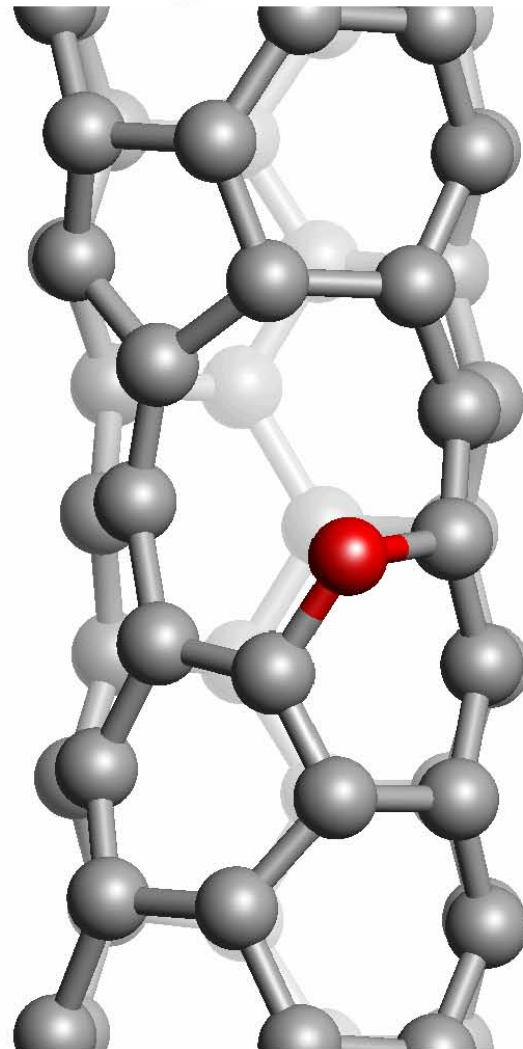
Electronic excitations!



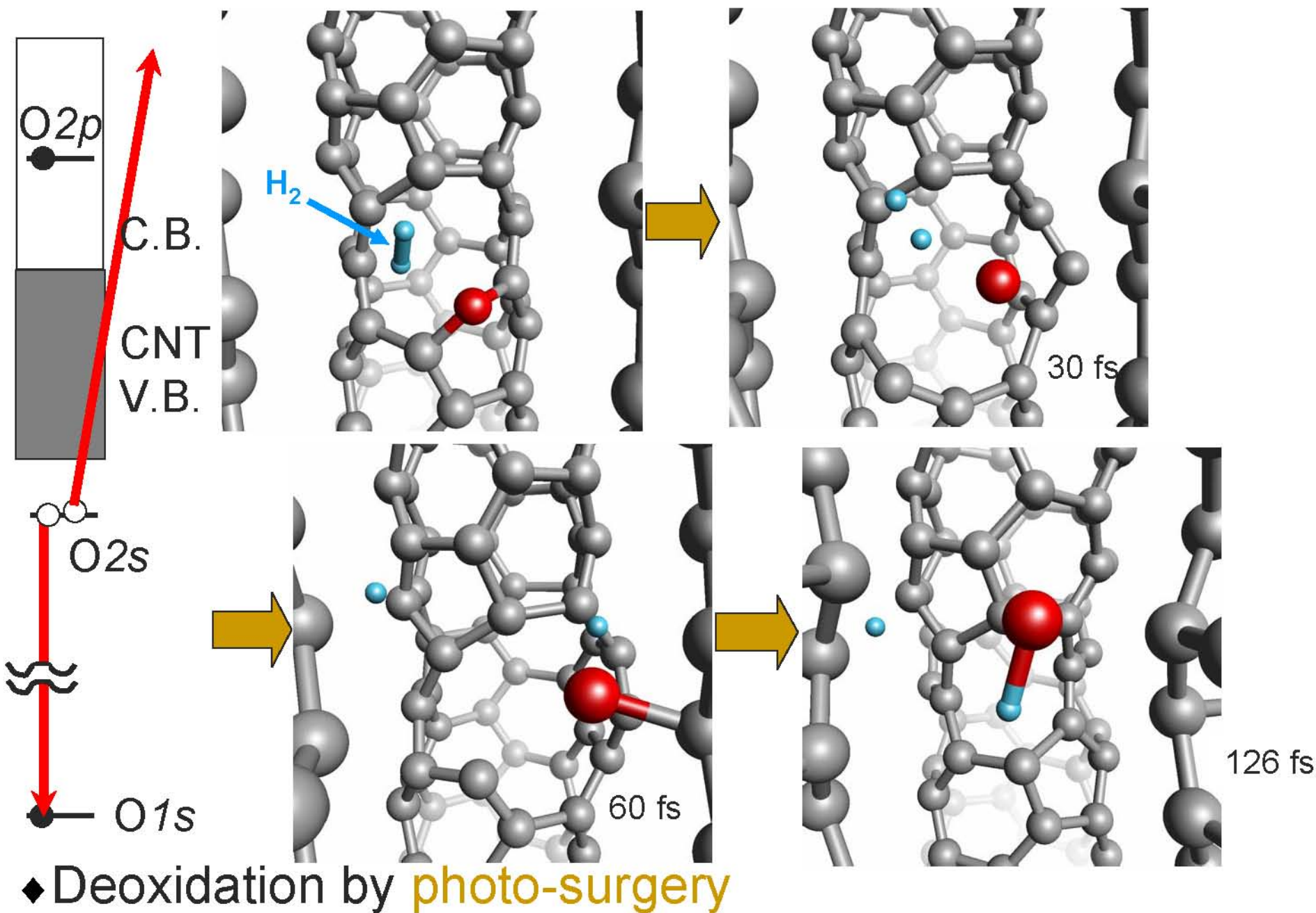
$O2s \rightarrow O2p$ excitation (33 eV)



hopeless



Auger decay following the $O1s \rightarrow 2p$ excitation (~ 520 eV)



Summary and Conclusions

- **Fusion of fullerenes** inside a nanotube starts with a cycloaddition and continues exclusively with Stone-Wales transformations.
- **Fusion of nanotubes** occurs efficiently via a zipper mechanism.
- Carbon nanotubes are Nature's best **thermal conductors**.
- Photo-excitations may be used to **detect specific defects** by their vibrational signature.
- **Hot carriers decay** by electron-electron, subsequently by electron-phonon scattering.
- Heat and photo-excitations may induce **self-healing** behavior in defective nanotubes.
- Photo-excitations can be used to **selectively remove oxygen** impurities.

The End