

Theoretical Spectroscopy and Electronic Excitations

Arvid Ihrig, Fabio Caruso^{*} and Patrick Rinke

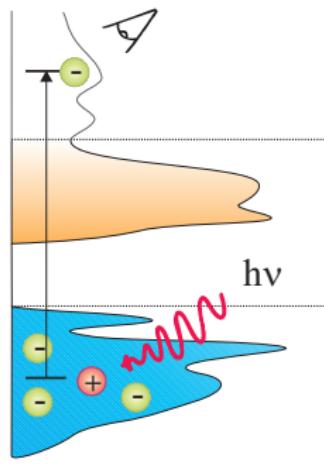
Fritz-Haber-Institut der Max-Planck-Gesellschaft, Berlin

^{*} Department of Materials, University of Oxford, Oxford

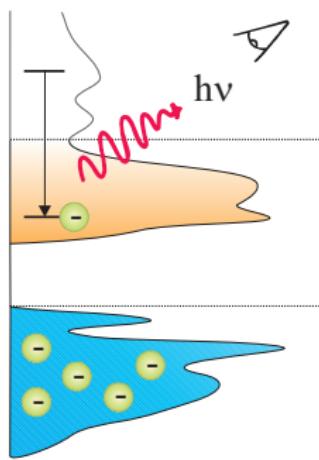
Electronic Structure Theory for Materials and (Bio)molecules
Tutorial 5 [Los Angeles 2014]

Band structures: photo-electron spectroscopy

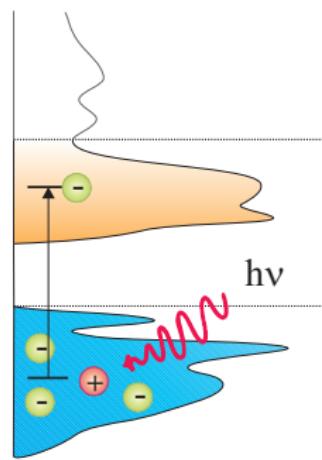
Photoemission



Inverse Photoemission



Absorption

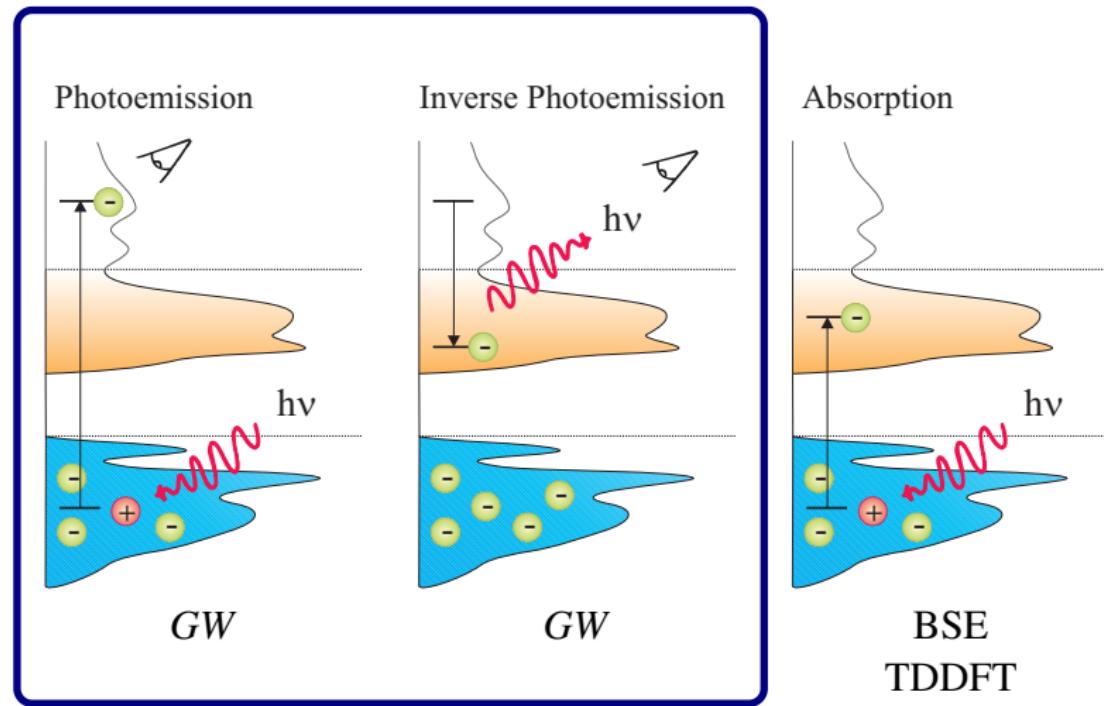


GW

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BSE
TDDFT

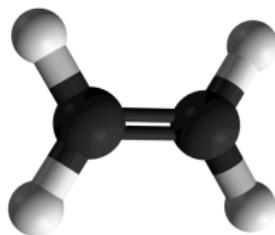
Band structures: photo-electron spectroscopy



In this tutorial: Theoretical spectroscopy of C₂H₄

Ethylene:

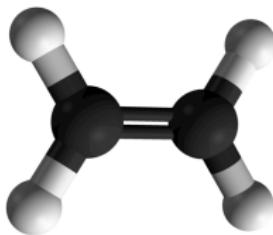
- simplest unsaturated hydrocarbon (after acetylene (C₂H₂)))



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Ethylene:

- simplest unsaturated hydrocarbon (after acetylene (C₂H₂)))
- key component of polyethylene (used for plastics of any shape and form)
- 109 million tonnes produced worldwide (in 2006)



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Ethylene:

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- plant hormone: induces fruit ripening

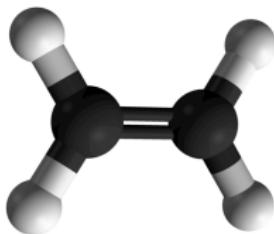


Photo-Electron Excitation Energies (Work in progress)

- Photoemission

- ▶ electron removal
- ▶ removal energy
 $E(N)$

 $|N\rangle$

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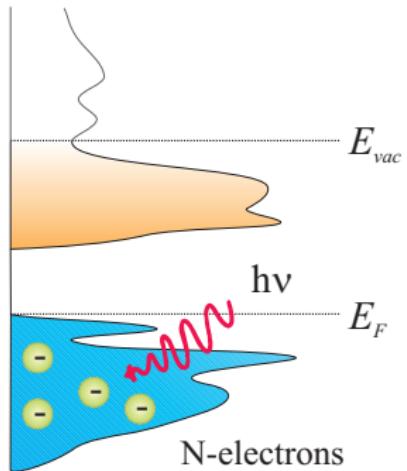


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$$\hat{\psi}(\mathbf{r}) |N\rangle$$

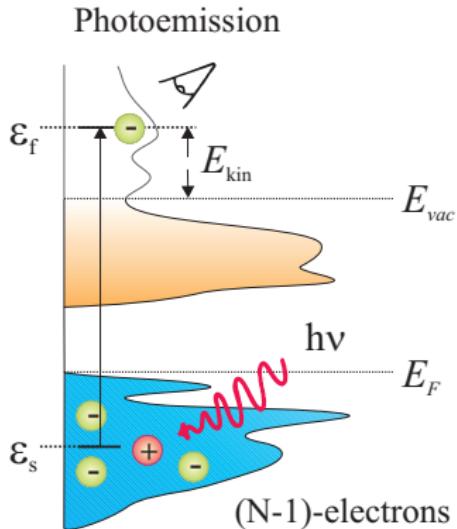


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$$E(N) - E(N - 1, s)$$

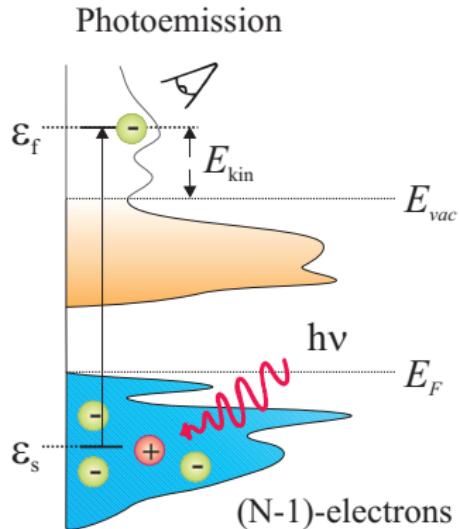


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$$\epsilon_s = E(N) - E(N - 1, s)$$

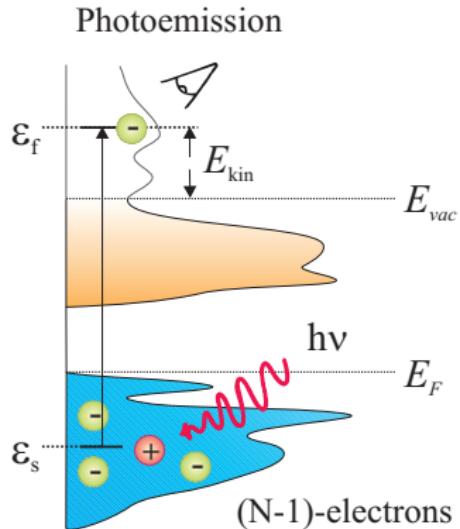


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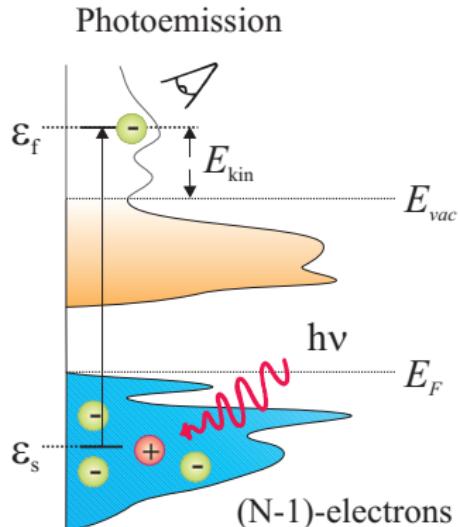
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- ▶ electron addition

$$\psi_s(\mathbf{r}) = \langle N + 1, s | \hat{\psi}(\mathbf{r})^\dagger | N \rangle$$

- ▶ addition energy

$$\epsilon_s = E(N + 1, s) - E(N)$$



Ionisation Potential, Electron Affinity and (Band) Gaps

We could use a total energy method to compute excitation energies:

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Band Gap energy difference between HOMO and LUMO

$$E_{gap} = I - A$$

What about Hartree-Fock eigenvalues?

Koopmans' theorem (Physica 1, 104 (1934))

$$E^*(N \pm 1, s) - E(N) = -\epsilon_s^{\text{HF}}$$

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Problems with this approach:

- orbital relaxations can be important
- correlations absent from HF

Janak's theorem (PRA 18, 7165 (1978)))

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rearranging and making mid point approximation:

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In exact KS-DFT the ionization potential is given by the
KS eigenvalue of highest occupied state.

In this tutorial (Part I)

In the first part of this tutorial, you will

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- calculate the ionization potential using the HF eigenvalue
- compute the ionization potential and electron affinity using the Δ -SCF method

Δ -SCF method

- compute the DFT ground state energy of the neutral system
- compute the DFT ground state energy of the charged system
- compute the ionization potential / electron affinity as the energy difference between the two charge states

Single Particle Green's Functions

- single particle Green's function G

$$iG(\mathbf{r}, \mathbf{r}'; t - t') = \left\langle \psi_0^N \left| \hat{T} \left[\psi(\mathbf{r}, t) \psi(\mathbf{r}', t')^\dagger \right] \right| \psi_0^N \right\rangle$$

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- Lehmann Representation of the Green's function:

$$\begin{aligned} iG(\mathbf{r}, \mathbf{r}'; \omega) &= \sum_k \frac{\langle \Psi_0^N | \psi(\mathbf{r}) | \Psi_k^{N+1} \rangle \langle \Psi_k^{N+1} | \psi(\mathbf{r})^\dagger | \Psi_0^N \rangle}{\omega - (E_k^{N+1} - E_0^N) + i\eta} \\ &\quad + \sum_k \frac{\langle \Psi_0^N | \psi(\mathbf{r})^\dagger | \Psi_k^{N-1} \rangle \langle \Psi_k^{N-1} | \psi(\mathbf{r}) | \Psi_0^N \rangle}{\omega + (E_k^{N-1} - E_0^N) - i\eta} \end{aligned}$$

Single Particle Green's Functions

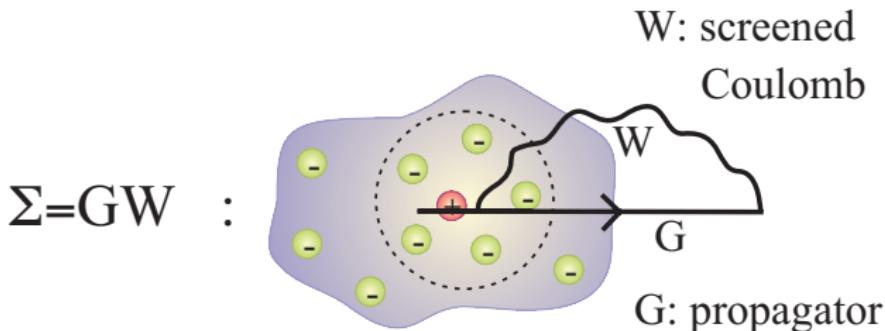
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GW Approximation - Screened Electrons

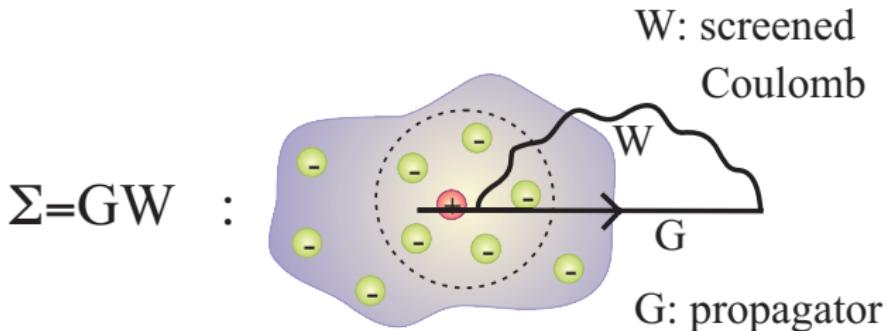


Self-Energy

$$\Sigma^{GW}(\mathbf{r}, \mathbf{r}', \omega) = -\frac{i}{2\pi} \int d\omega e^{i\omega\eta} G(\mathbf{r}, \mathbf{r}', \omega + \omega') W(\mathbf{r}, \mathbf{r}', \omega')$$

$$\epsilon_s^{qp} = \epsilon_s^{\text{KS}} + \langle s | \Sigma(\epsilon_s^{qp}) | s \rangle - \langle s | v_{xc} | s \rangle$$

GW Approximation - Screened Electrons



Self-Energy: $\Sigma = \Sigma_x + \Sigma_c$

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$\Sigma_x = iGv$ exact (Hartree-Fock) exchange

$\Sigma_c = iG(W - v)$ correlation (screening due to other electrons)

GW -- How to obtain the self energy

Hedin's GW equations:

$$G(1, 2) = G_0(1, 2) \quad \text{notation: } 1 = (\mathbf{r}_1, \sigma_1, t_1)$$

$$\Gamma(1, 2, 3) = \delta(1, 2)\delta(1, 3)$$

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Convergence of GW -methods

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Dyson's equation:

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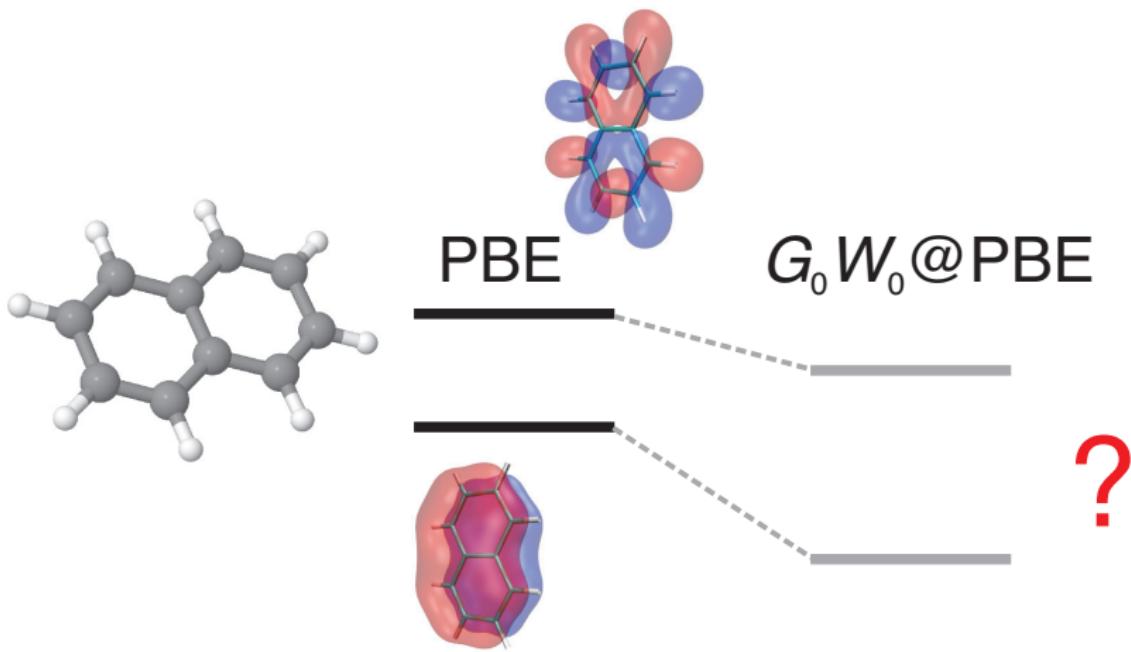
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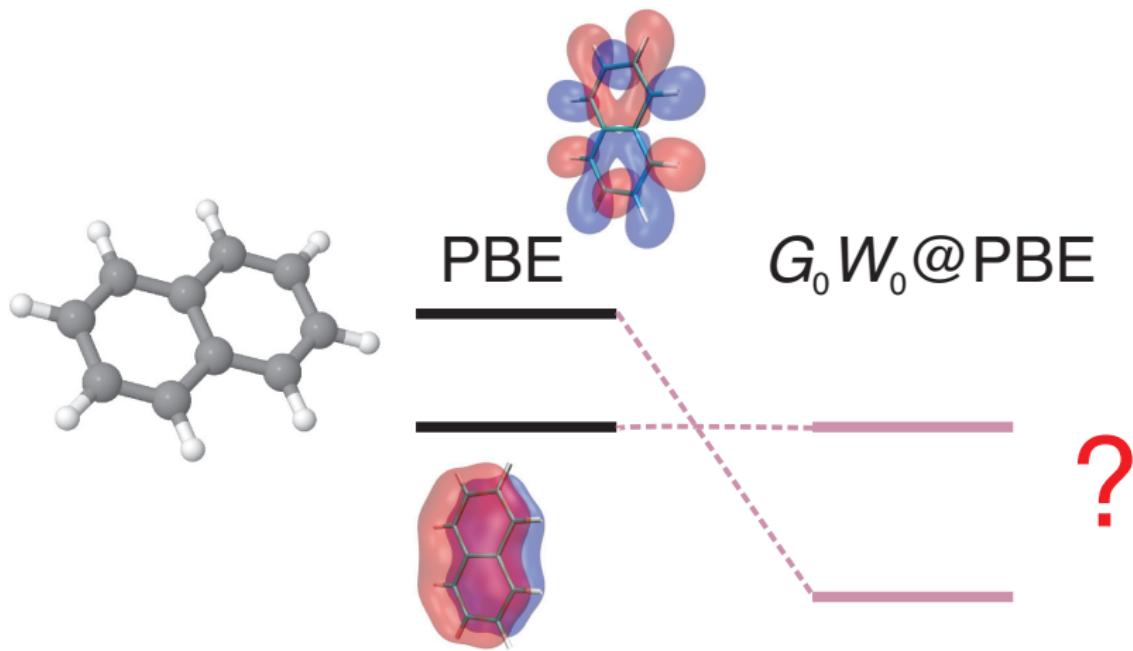
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- check the starting point independence of sc GW

Naphthalene and the self-interaction error



Naphthalene and the self-interaction error



Ground state properties from Green's functions

- The Green's Function also gives access to ground state properties.
- Galitskii-Migdal formula for total energy

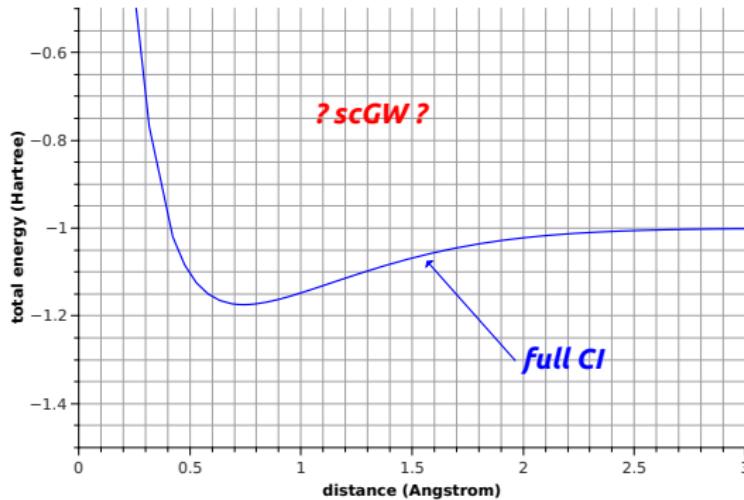
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- In this tutorial: application to H₂ binding curve



Your tutors this afternoon



Patrick Rinke



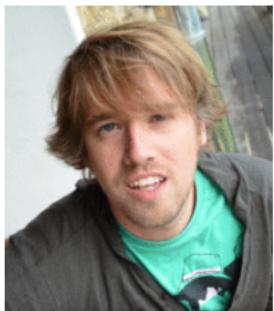
Arvid Ihrig



Björn Bienek



Xinguo Ren



Daniel Berger

Your tutors this evening



Oliver
Hofmann



Mariana Rossi